



Occurrence of structural aluminium (Al) in marine diatom biological silica: Visible evidence from microscopic analysis

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Abstract.

The global marine biogeochemical cycle of aluminum (Al) is believed to be driven by marine diatoms, due to the uptake of dissolved Al (DAI) by living diatoms from surface seawater. The occurrence of Al in diatom biogenic silica (BSi) can inhibit the dissolution of BSi, thus benefiting the effects of the ballast role of diatoms in the biological pump and forming a coupled Si-Al biogeochemical cycle. However, the occurrence mechanism of Al in marine diatoms is still unclear. In particular, whether or not Al is incorporated into the structure of BSi of living diatoms is unrevealed, resulting in difficulties in understanding the biogeochemical behaviors of Al. In this study, *Thalassiosira weissflogii*, a widely distributed marine diatom in marginal seas, was selected as the model to evaluate the occurrence of structural Al in BSi based on culturing experiments with the addition of DAI. The structural Al in BSi was detected by combining focused ion beam (FIB) scanning electron microscopy and energy dispersive X-ray spectroscopy (EDS) mapping analysis. Direct evidence of structural Al in living BSi was obtained for the first time. The distribution and content of this Al were revealed by the EDS-mapping analysis. The structural Al in the BSi exhibited a homogeneous distribution, and the average Al/Si atomic ratio obtained through the FIB-EDS mapping analysis was 0.011. The effects of structural Al on BSi dissolution-inhibition are discussed based on the content of this Al. The fundamental results indicate the significant contribution of marine diatoms to the biogeochemical migration of marine Al.

1 Introduction

Aluminum (Al) is the most abundant metal element in the Earth's crust, and it mainly occurs in Al-bearing minerals such as aluminosilicate (Taylor, 1964). The weathering of such aluminosilicates results in a huge carbon (C) sink in rocks and soils



35 due to the transformation of CO_2 to HCO_3^- based on the reaction with such Al-bearing minerals (Mackenzie and Kump, 1995; Wallmann *et al.*, 2008). Therefore, Al is involved in the terrestrial C cycle (Stallard, 1998). Despite the high-Al content of the terrestrial system, the concentration of dissolved Al (DAI) in natural water is very low (from $<$ tens of nM in oceans to μM in freshwater) (Hydes, 1977; Measures and Edmond, 1990; Chou and Wollast, 1997; Menzel *et al.*, 2020), due to the low dissolution of Al-bearing minerals; and thus, oceanic DAI is often used as a tracer for the transportation of terrestrial

40 substance and the mixing of different water masses (Measures and Vink, 1999; 2000; Han *et al.*, 2008; Schlitzer *et al.*, 2018). Due to its lack of an essential biological role (even toxicity) (Xie *et al.*, 2015; Gillmore *et al.*, 2016), the terrestrial cycling of Al is believed to be primarily controlled by inorganic processes. However, marine diatoms are found to take up and incorporate Al into their cells (Q Liu *et al.*, 2019). Diatoms can eliminate plenty of DAI from surface water when they bloom (Hall *et al.*, 1999; J-L Ren *et al.*, 2011), and they export Al into the sedimentary layers through the sedimentation of post-

45 mortem diatoms, forming a coupled Si-Al biogeochemical cycle in the oceans (van Hulst *et al.*, 2013; 2014). Diatoms are a type of widely distributed single-cell algae in the oceans, and they account for up to $\sim 40\%$ of the oceanic primary production (Armburst, 2009). Through sedimentation of dead diatoms, huge amounts of organic carbon (OC) are exported from the surface ocean to the deep ocean (Nelson *et al.*, 1996; Riebesell, 2000; Leblanc *et al.*, 2018) controlling the oceanic C cycle and forming an important part of the biological carbon pump (Smetacek, 1999; Tréguer and Pondaven, 2000;

50 Ragueneau *et al.*, 2006). Compared with other marine phytoplankton, diatoms are more effective at carbon sequestration over longer timescale due to their frustules, which are composed of biogenic silica (BSi) with a high-mechanical stability. Since it acts as ballast (De La Rocha *et al.*, 2008; Honda and Watanabe, 2010), BSi carries the OC to deeper oceans and deposits on the seafloor, forming a coupled Si-C cycle.

55 Considering the key role of diatoms in carbon sequestration, the Al in diatom cells is also involved in the C cycle in the oceans, and thus, Al participates in the global C geochemical cycle through inorganic and biological processes (Gehlen *et al.*, 2003; H Ren *et al.*, 2013). Although preliminary studies have been carried out to investigate the Al in freshwater (D Liu *et al.*, 2019) and marine diatoms, the occurrence of Al in marine diatoms is far away unclear, which results in difficulties in understanding the abovementioned diatom-derived Al cycle.

60 It has been proposed that Al is incorporated into the organic components and BSi of diatoms. For the former, the occurrence mechanism and distribution have been well described (Q Liu *et al.*, 2019). However, for the latter, whether Al occurs in the structure of BSi is ambiguous. Many researches have tried to prove the existence of structural Al based on the detection of BSi sourced from living marine diatoms and diatom fossils by removing the organic component of the diatoms and the Al-bearing matter on the surface, respectively (Gehlen *et al.*, 2002; Koning *et al.*, 2007; Machill *et al.*, 2013). However, Al-

65 bearing tiny mineral particles can be hardly excluded from sedimentary BSi even with the greatest of care, e.g., the acid washing reported by Gelhen (Gehlen *et al.*, 2002). Moreover, the high activity of fresh BSi results in the easy adsorption of DAI from Al-contained solutions during the removal of the Al-bearing organic components of diatoms, which interferes with the evaluation of the structural Al (Moran and Moore, 1988; Koning *et al.*, 2007). Due to these challenges, the occurrence of



Al in the BSi structure still unclear, although possible contents and coordinated states have been proposed. The reason that
70 so much attention has been paid to the structural Al in BSi is that structural Al has a dissolution-inhibition effect on BSi.
About 25% decrease in the solubility of BSi was speculated when Al substitutes for 1 out of every 70 Si atoms (*Dixit et al.*,
2001), and thus, structural Al is believed to be one of the key factors that influence the migration of BSi from the surface
ocean to pelagic sediments. Therefore, the occurrence of Al in BSi is indefinite, which severely limits our understanding of
the composition, structure, and water stability of BSi.

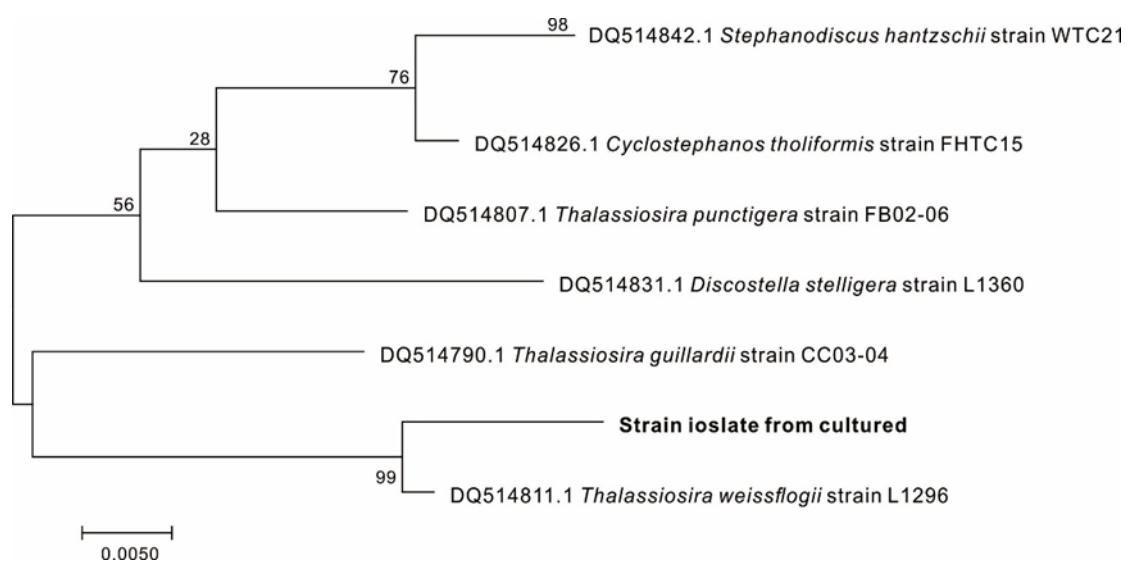
75 In this study, a widely distributed marine diatom in marginal seas, *Thalassiosira weissflogii* (*T. weissflogii*)
(*Panagiotopoulos et al.*, 2020), was selected as the model; and culture experiments were performed with the addition of high
concentrations of DA1. The direct detection of the internal surface of BSi was conducted using a focused ion beam (FIB)
pretreatment, which can remove the external surface, leading to purification of BSi. Energy dispersive X-ray spectroscopy
(EDS) mapping analysis was used to determine the content and distribution of the structural Al. Finally, the coupled Si-Al
80 cycle controlled by marine diatoms was discussed.



2 Materials and Methods

2.1 Cultivation of the diatoms

85 *T. weissflogii*, purchased from the Shanghai Guangyu Biological Technology Co., Ltd (China), was used as the model diatom in this study to investigate Al occurrence. The species of the marine diatom samples was identified by determining their *rbcL* gene sequences (Couradeau *et al.*, 2012). The obtained phylogram clearly showed that the diatoms obtained was *T. weissflogii* (Fig. 1).



90 **Fig. 1. Phylogenetic trees of diatom strains isolated from the culturing diatoms. The scale bar indicates the number of substitutions per site for a unit branch length.**

The diatoms were cultured in artificial seawater supplemented with f/2 medium at 25°C under a 12/12 light/dark cycle at an intensity of 100 mol photons m⁻²s⁻² for 20 days. The composition of artificial seawater and f/2 culture medium are presented in Table. S1 (see Supporting Information). Deionized water was used to prepare the artificial seawater and f/2 medium. Cell of the diatoms in the culture media with/without the addition of Al was counted using flow cytometry (BioMarine-IA 1000, Countstar Company, Shanghai, China).

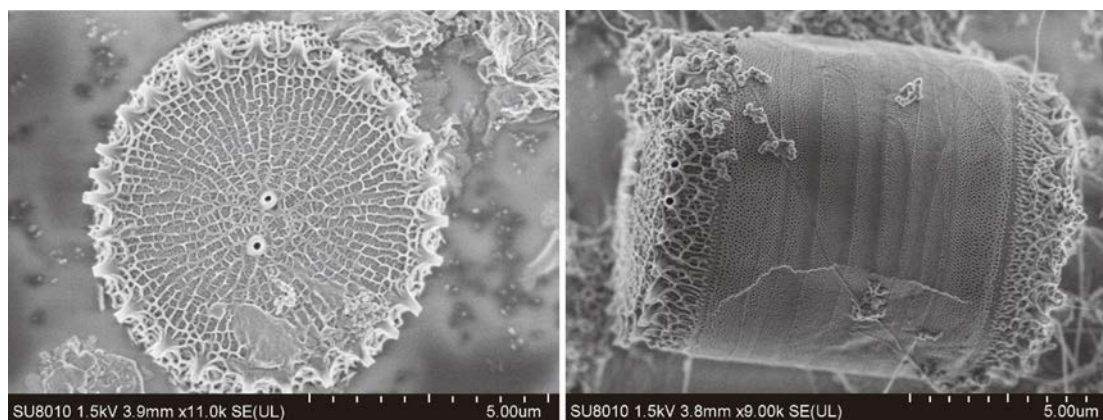
100 Trace-metal clean processes were used to eliminate the interference from other elements. Before the cultivation experiment, all of the bottles and tubes were soaked in 10% hydrochloric acid for 24 h, sterilized at 120°C for 30 min, and washed repeatedly with deionized water (Q Liu *et al.*, 2019).

AlCl₃ (i.e., the Al source) was added to the culture medium with an initial concentration of 2.0 μM in the medium. The control experiment was also simultaneously performed without adding AlCl₃. The pH value of the diatom culture medium was adjusted to 8.0 using NaOH solution. All of the chemical reagents were allowed to stand overnight to reach chemical equilibrium before use.



105 2.2 Collection of diatoms and extraction of diatom frustules

After 14 days of culturing, the diatoms were collected through high-speed centrifugation at 11000 rpm. The obtained diatoms (Fig. 2) were rinsed with deionized water three times to remove any impurities adsorbed on the surface. The solid obtained was freeze-dried after centrifugation. Two pre-treatment procedures were used to remove any Al adsorbed on the surface of the diatoms and their organic components in order to obtain pure BSi: 1) immersion in 0.05 M EDTA for 12 h
110 followed by washing three times; and 2) immersion in 30% hydrogen peroxide solution for 48 h followed by washing five times. The removal of the organic components was confirmed using a Vario EL III elemental analyzer with a detection limit of <40 ppm. Approximately 2.00 mg of sample was placed in a 6 × 12 mm tin boat and was oxidized in a combustion tube in the presence of oxygen at high temperature ($\geq 1150^{\circ}\text{C}$).



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Fig. 2. Scanning electron microscopy (SEM) of *T. weissflogii*

2.3 Characterization and Data analysis

2.3.1 Focused ion beam (FIB) treatment and characterization methods

120 The FIB thinning was carried out using FIB scanning electron microscopy (SEM; FEI Helios Nano Lab 450S) equipped with a flip stage, an in situ scanning transmission electron microscope (STEM) detector, a Tomahawk ion column, and a multichannel gas injection system. For the FIB milling conducted to obtain a thin BSi slice, a 5 kV focused gallium ion (Ga^+) beam with a beam current of 40 pA was used, and the total thinning time was 2 min. Then, the obtained BSi slices were fixed on the edge of the FIB half grid through induced platinum (Pt) deposition. The in situ field emission SEM (FE-SEM)
125 observation of the BSi slice was performed with an accelerating voltage of 30 kV and a current of 24–9300 pA. The elemental distribution of the sliced BSi was obtained using an energy-dispersive X-ray spectrometer attached to a transmission electron microscope (FEI Talos F200 TEM/EDS microscope) with a voltage of 200 kV.

2.3.2 Energy dispersive spectroscopy (EDS) spot analysis

In previous studies, the change in the Al concentrations of the culture medium was used to evaluate the Al uptake of diatoms.
130 However, some of the Al form precipitated due to changes in the chemical environment, rather than being involved in the



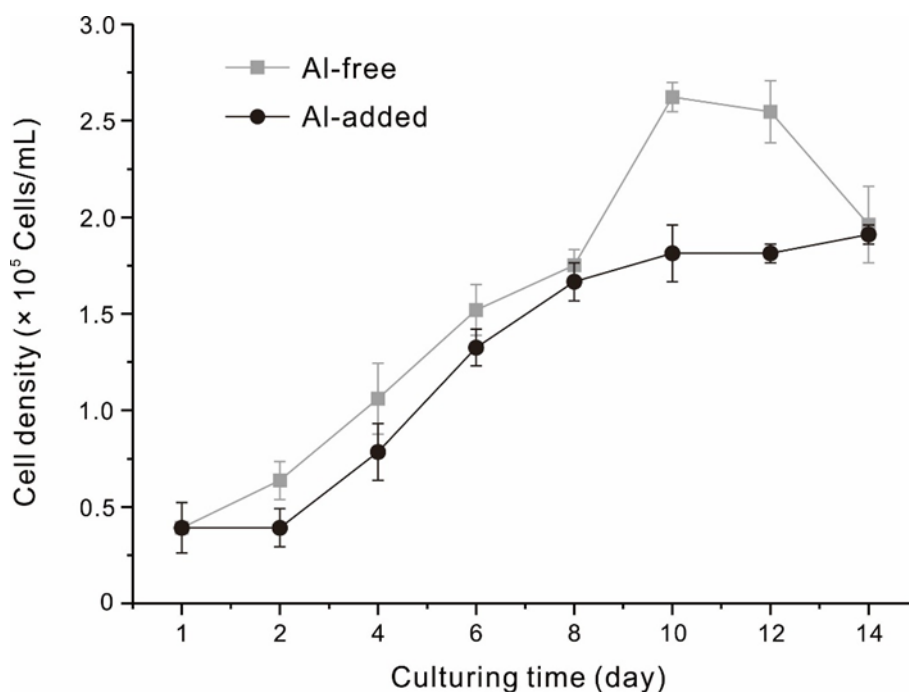
biological processes of diatoms. To avoid the disturbance of such Al, which does not incorporate into the cells of the diatoms, the Al/Si atomic ratios of the cultured diatoms and their BSi were determined using EDS spot analysis and a scanning electron microscope (SU8010, UltraPore-300, PDP-200). A total of 353 spots from 35 whole diatoms and the BSi of the *T. weissflogii* were detected, and the Al/Si atomic ratios were obtained at a voltage of 15 kV and a current of 20 μ A. Carbon coating was performed before the SEM-EDS analysis.

The data for the specific growth rate and EDS spot analysis were obtained on the basis of the mean \pm standard deviation (S.D.; $n \geq 3$). One-way analysis of variance (ANOVA) was used to determine the significant differences ($p < 0.05$) between the treatments (Shi *et al.*, 2015).

140 3 Result

3.1 Effects of Al on the growth of diatoms

To investigate the influence of Al on the growth of *T. weissflogii*, the cell density was measured during the diatom cultivation processes. The stable growth period of the diatoms cultured with and without the addition of Al began on the third day. The growth trends of the diatom cells were similar in the diatom culture mediums with and without the addition of Al (Fig. 3). The final concentration of the diatom cells exhibited a very slight difference, with average values of 1.96×10^5 and 1.91×10^5 cells/mL for the Al-free diatom culture medium and the Al-added medium, respectively. The results indicate that the presence of Al did not significantly affect the cell yields of *T. weissflogii* during the 14-day culture period.





150 **Fig. 3. The cell densities of the diatoms cultured in culture mediums with (Al-added) and without (Al-free) the addition of Al during the 14-day culturing period.**

3.2 Identification of structural Al in BSi based on FIB-EDS

155 FIB milling can remove the external surface of the diatom and impurities adsorbed onto it, allowing for the detection of the inner structure of the BSi. Thus, the subsequent EDS detection obtains the structural element distributions of the BSi. In this case, the combined FIB–EDS analyses avoided the disturbance of the non-structural elements. The *T. weissflogii* BSi sample was selected (Fig. 4a and the inset) and milled to a thin slice using FIB (Figs. 4b and c). Two inner areas of the BSi slice were observed (blue rectangles in Fig. 4d), and the Al distribution was detected in these two domains, revealing a homogeneous distribution of Al accompanied by Si in the BSi structure (Figs. 4e and g).

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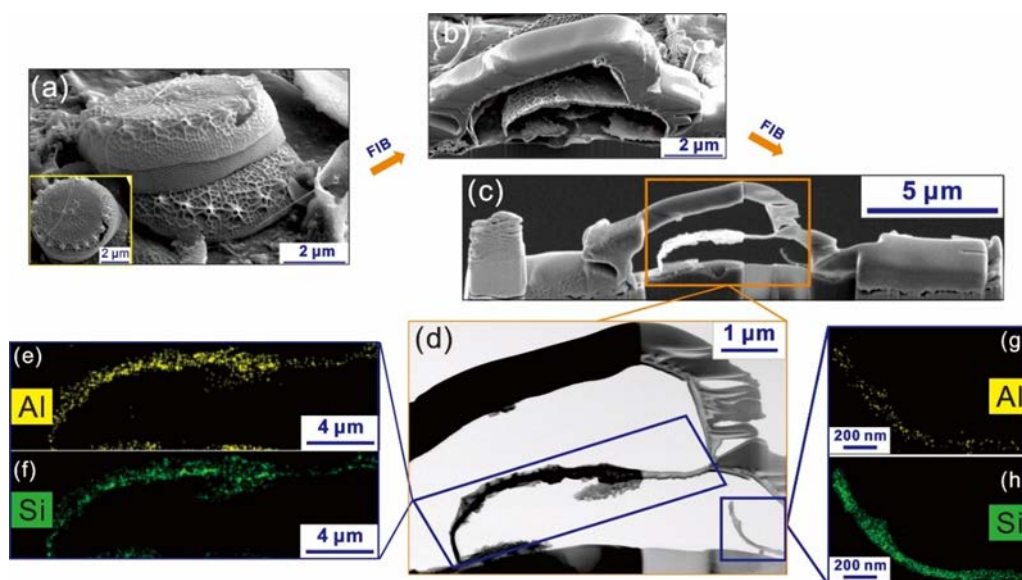


Fig. 4. (a and the inset) FESEM image of a *T. weissflogii* BSi; (b and c) FESEM images of a series of FIB thinning products of the BSi during the ion sputtering experiments; (d) HRTEM bright field (BF) image of the selected domain in the rectangle in (c); (e and g) Al and (f and h) Si distributions in the domains in (d) obtained using EDS mapping analysis.

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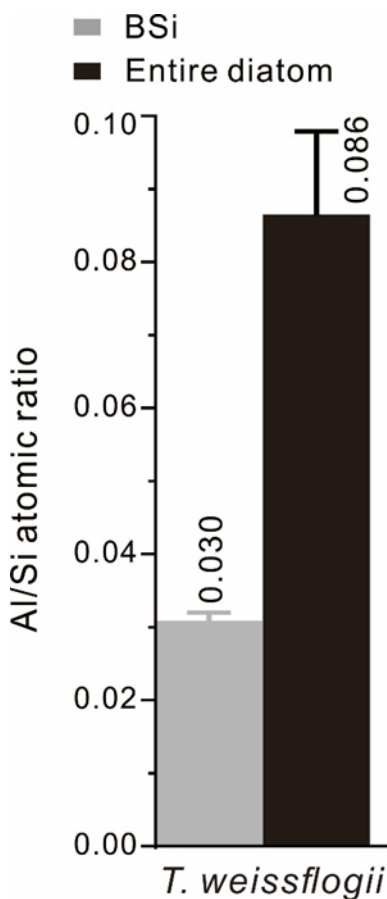
This is direct evidence of the presence of Al in the internal surface of BSi sourced from living marine diatoms, demonstrating the occurrence of structural Al in the diatomaceous biological framework of marine diatoms. The Al is incorporated into the BSi based on the biological behaviors of marine diatoms and is used to build diatom BSi. Based on the EDS mapping analysis, the average Al/Si atomic ratio is 0.011.

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3.3 Quantity measurement of the Al in the diatoms and their BSi using EDS spot analysis



175 More than 300 spots on diatoms and their BSi were analyzed using EDS, and the average Al/Si atomic ratios were 0.086 and 0.030 for the entire diatoms and their BSi, respectively. The Al/Si value of the BSi is lower than that of the entire diatoms (Fig. 5), because Al not only occurs in BSi but also in the organic components of the diatoms. The Al content of the organic component is nearly 3-fold higher than that of the BSi. Moreover, the average Al/Si atomic ratio of the BSi (0.030) obtained through EDS spot analysis is higher than that obtained through FIB-EDS mapping (0.010).



180 **Fig. 5** Al/Si atomic ratios of the *T. weissflogii* diatom cells (the black column) cultured in media with the addition of Al and of their BSi (the grey column). The EDS spot analysis used to determine the Al/Si atomic ratios was based on 353 spots on 35 diatom cells and their BSi samples. The error bars were obtained through mean \pm standard deviation analysis of all of the Al/Si atomic ratios

4 Discussion

185 The scavenging of Al from seawater by diatoms has been widely observed during diatom blooms (*Hall et al.*, 1999; *J-L Ren et al.*, 2011). This unique biological behavior of Al uptake has attracted a great deal of attention, and a corresponding biogeochemical Al cycle driven by diatoms was hypothesized (*van Hulst et al.*, 2014). Proving the occurrence of Al in



190 diatoms is key to understanding the mechanism of Al uptake by marine diatoms. Zhou *et al.* investigated Al incorporation into diatoms and proposed the distribution and the content of Al in the organic components of diatoms based on various extraction methods (Q Liu *et al.*, 2019). However, the occurrence of Al in BSi of marine diatoms was still unknown (Gehlen *et al.*, 2002). Al-bearing impurities, which are hardly removed by physical and chemical pre-treatments, influence the detection of the Al incorporated in the BSi, preventing the identification and quantification of the structural Al. Moreover, Al-bearing minerals have been found to form on the surfaces of BSi after diatoms are dead. Due to these problems, the Al detected through elemental analyses such as X-ray fluorescence (XRF), inductively coupled plasma optical emission spectroscopy (ICP-OES) for the DAl in diatom culture medium (Machill *et al.*, 2013), and EDS for marine BSi is unable to
195 confirm the presence of Al in BSi. Therefore, until now, the lack of direct evidence of Al in BSi has resulted in the existence of structural Al sourced from biological uptake being ambiguous.

T. weissflogii shows a high Al tolerance, and its growth is not influenced by Al at such a high initial Al concentration (2.0 μM). These results are in good agreement with the results of some previous reports (D Liu *et al.*, 2019). This high Al tolerance may be due to the biological adaptation of this diatom, which lives in estuary and continental shelf environments
200 with high environmental Al concentrations (Zhou *et al.*, 2018).

In our previous study, we found that FIB milling allowed to obtain a slice of an internal section of BSi and to remove the external surface, avoiding the disturbance of any impurities in the further detection of the structural elements (D Liu *et al.*, 2019). The freshwater BSi was investigated to reveal Al occurring in the structure of BSi and the high level concentration of Al was detected. However, whether Al is incorporated into the silica structure of marine diatoms or not is still an opening
205 question. Gehlen *et al.* (2002) and Koning *et al.* (2017) proposed the occurrence of structural Al in marine BSi based on the Al K-edge X-ray absorption near-edge structure (XANES) spectra. However, as declared by Gehlen *et al.* (2002), the detection was impossible to avoid the interference of the Al-bearing phase. Through a combination of FIB and EDS, direct and visible evidence was obtained, illustrating the distribution and quantity of the structural Al in BSi. For the first time, structural Al was observed in BSi of marine diatoms, demonstrating that marine diatoms take up Al and use it to build their
210 siliceous framework (BSi). Similar to freshwater diatom BSi, Al is homogeneously distributed in marine BSi, but the Al content of marine diatoms (Al/Si atomic ratio of ~ 0.011) is much lower than that of freshwater diatoms (~ 0.050) based on the Al/Si atomic ratio (D Liu *et al.*, 2019; Yuan *et al.*, 2019). However, the value is slightly higher than the maximum (0.008) reported in previous studies (Gehlen *et al.*, 2002; Koning *et al.*, 2007). This may be due to the high Al concentration of the culture medium and the preference of the diatoms for Al. It should be noted that the Al/Si atomic ratio of BSi, which was
215 extracted from living diatoms by removing the organic components of the diatoms, is much larger than the value obtained by FIB-EDS. This result may be due to the disturbance of non-structural Al, which is sourced from the deconstruction of the organic components of the diatoms and is subsequently adsorbed onto the surface of BSi.

Al is incorporated into both the organic components and BSi of diatoms after being taken up through biological behaviors. The Al content of the organic components is higher than that of BSi, indicating that the contribution of the organic
220 components to the Al occurrence in diatoms is greater than that of the BSi. However, considering the fact that the organic



components start to be degraded by marine bacteria as soon as the diatoms die (*Bidle and Azam, 1999*), the Al in the organic components may return to seawater, which accounts for ~75% of the total Al in the diatoms, and thus only the smaller proportion of the Al that occurs in the BSi structure may be deposited with the BSi. The BSi in sediments constitutes a diatom-driven biological Al sink.

225 Moreover, structural Al is proposed to have a dissolution-inhibition effect on BSi, i.e., a 25% decrease in BSi solubility when Al substitutes for 1 out of every 70 Si atoms (Al/Si atomic ratio of 0.014) (*Dixit et al., 2001*). Since the Al/Si atomic ratio of the BSi structure is 0.11, the BSi solubility will be decreased by ~20%, based on the relationship between Al content and BSi dissolution-inhibition. Therefore, this structural Al makes a high contribution to BSi burial. Considering that a huge quantity of carbon settles into the sedimentary layers every year based on BSi exports (*Smetacek, 1999; Tréguer et al., 2017*),
230 the dissolution-inhibition effect of structural Al on BSi may influence or even control the efficiency of the carbon sequestration of the diatom-driven biological pump.

5 Conclusions

In this study, the occurrence of structural Al in BSi of marine diatoms was investigated using a combination of FIB
235 pretreatment and SEM-EDS analyses. BSi from living *T. weissflogii* diatoms was obtained from culture mediums with an Al concentration of 2.0 μM , indicating that Al is not toxic to *T. weissflogii*. Direct, visible evidence of the presence of structural Al in living BSi was obtained for the first time, and the structural Al was found to have a homogeneous distribution within the silica framework based on the FIB-EDS mapping analysis of BSi slices, the average Al/Si atomic ratio in the structure of the BSi was 0.011, implying that the structural Al makes a high contribution to the dissolution-inhibition of the BSi.
240 Considering the influence of structural Al on the solubility of diatom biosilica, the occurrence of structural Al in BSi can influence the carbon sequestration efficiency of the diatom-driven biological pump.

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