



Expanding seawater carbon dioxide and methane measuring capabilities with a Seaglider

Claudine Hauri¹, Brita Irving¹, Dan Hayes², Ehsan Abdi^{3,a}, Jöran Kemme⁴, Nadja Kinski⁴, and Andrew M. P. McDonnell^{5,b}

¹International Arctic Research Center, University of Alaska Fairbanks, Fairbanks, AK 99775, USA

²Advanced Offshore Operations, Inc., Houston, TX 77004, USA

³Cyprus Subsea Consulting and Services, Lakatamia 2326, Cyprus

⁴-4H-JENA engineering GmbH, 07745 Jena, Germany

⁵College of Fisheries and Ocean Science, University of Alaska Fairbanks, Fairbanks, AK 99775, USA

^anow at: Akvaplan-Niva, 9296 Tromsø, Norway

^bnow at: Alaska Renewables, Fairbanks, AK 99709, USA

Correspondence: Claudine Hauri (chauri@alaska.edu)

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Abstract. Warming, ocean acidification, and deoxygenation are increasingly putting pressure on marine ecosystems. At the same time, thawing permafrost and decomposing hydrates in Arctic shelf seas may release large amounts of methane (CH₄) into the water column, which could accelerate local ocean acidification and contribute to climate change. The key parameters to observing and understanding these complex processes and feedback mechanisms are vastly undersampled throughout the oceans. We developed carbon dioxide (CO₂) and CH₄ gliders, including standard operational procedures, with the goal that CO₂ and CH₄ measurements will become more common for glider operations. The Seagliders with integrated Contros HydroC CO₂ or CH₄ sensors also include conductivity, temperature, depth, oxygen, chlorophyll *a*, backscatter, and fluorescent dissolved organic matter sensors. Communication via satellite allows for near-real-time data transmission, sensor adjustments, and adaptive sampling. Several sea trials with the CO₂ Seaglider in the Gulf of Alaska and data evaluation with discrete water and underway samples suggest nearly “weather-quality” CO₂ data as defined by the Global Ocean Acidification Network. A winter mission in Resurrection Bay, Alaska, provided the first insights into the water column inorganic carbon dynamics during this otherwise undersampled season. The CH₄ Seaglider passed its flight trials in Resurrection Bay but needs to be tested during a field mission in an area with

CH₄ concentrations beyond background noise. Both sensing systems are available to the science community through the industry partners (Advanced Offshore Operations and -4H-JENA engineering GmbH) of this project.

1 Introduction

Understanding the distribution and dynamics of carbon dioxide (CO₂) and methane (CH₄) in the ocean is crucial for predicting and mitigating climate change and ocean acidification impacts. Within the ocean, CO₂ levels (measured as the partial pressure of CO₂, *p*CO₂, and/or fugacity of CO₂) are spatially and temporally variable as they are influenced by a myriad of highly dynamic physical, chemical, and biological processes. On top of this natural variability, the ocean has absorbed about one-third of the CO₂ emitted by humans since the industrial revolution (Sabine et al., 2004; Gruber et al., 2019). In doing so, it has played an important role in mitigating climate change (Sabine and Tanhua, 2010). However, both the oceanic uptake of anthropogenic CO₂ and climate change are altering the distribution of oceanic CO₂ and are causing ocean acidification (Doney et al., 2009; Qi et al., 2022; Woosley and Millero, 2020). At the same time, the oceans are warming and losing oxygen (Johnson and Lyman, 2020; Breitbart et al., 2018), increasing the stress on marine

ecosystems. As these long-term changes unfold, marine heat waves and high-acidity or low-oxygen extreme events will last longer, become more intense, and happen more often and at the same time (Laufkötter et al., 2020; Gruber et al., 2021; Hauri et al., 2024). Negative effects on certain organisms are even stronger if exposed to a combination of different stressors (Breitberg et al., 2015; Kroeker et al., 2017).

Over the coming 100 years, CH₄ possesses a global warming potential approximately 28 times greater than that of CO₂ (IPCC AR5; Myhre et al., 2013). Sediments along the seafloor at continental margins contain large amounts of CH₄, with about 10 times as much carbon as the atmosphere (Kessler, 2014). CH₄ is biologically produced in anoxic sediments and the surface mixed layer or released from geological sources like hydrocarbon seeps and degrading methane hydrate deposits (Barnes and Goldberg, 1976; Du et al., 2014; Skarke et al., 2014). This powerful greenhouse gas is emitted to the atmosphere through bubbling (ebullition) or diffusive gas transfer (Reeburgh, 2007; McGinnis et al., 2006), which is limited by rapid oxidation to CO₂ during transport through the water column (Leonte et al., 2017). CH₄ generally occurs at low levels (background concentrations) throughout oceans unless close to a source. Positive feedback mechanisms, like warming-induced CH₄ seepage from destabilizing hydrates and thawing subsea permafrost, may further accelerate ocean acidification and climate change (Garcia-Tigreros et al., 2021; Sparrow et al., 2018; Shakhova et al., 2010; Rees et al., 2022).

To effectively observe and understand the complex processes and feedback mechanisms regulating Earth's systems, certain key parameters, defined by the Global Ocean Observing System as essential ocean variables, must be measured accurately. However, these variables are often vastly undersampled across time and space due to traditional sampling methods, which mainly rely on discrete water sample collections from dedicated research cruises, underway measurements from transiting vessels, or time series measurements from in situ sensors on fixed moorings. Although biogeochemical sensors deployed on autonomous platforms like moorings and Argo floats have become more prevalent, challenges such as high power requirements, sensor size, and data quality hinder their widespread use on underwater gliders. Autonomous, spatially resolved surface measurements of *p*CO₂ and pH are commonly collected using wave gliders and sail drones (Chavez et al., 2018; Nickford et al., 2022; Manley and Willcox, 2010). The state-of-the-art biogeochemical (BGC) Argo floats measure variables like pH, O₂, NO₃, chlorophyll *a*, suspended particles, and downwelling irradiance in subsurface waters (Claustre et al., 2020). These floats can last several years at low sampling resolutions, such as a 2000 m depth profile every 10 d, or they can be programmed for high-resolution and shallow sampling. They can even sample beneath seasonal sea ice (Briggs et al., 2018). Despite their capabilities, their trajectory is hard to control, and they are usually not recovered after their mis-

sion, which prevents sensor calibration and post-mission corrections.

Ocean gliders autonomously collect water column data along planned waypoints, which allows for controlled exploration and adaptive sampling. To date, pH is the only carbon system parameter that has been successfully integrated into ocean gliders (Hemming et al., 2017; Saba et al., 2019; Posenti et al., 2021; Takeshita et al., 2021). The most promising results came from ISFET-based pH sensors (ISFET: ion-sensitive field-effect transistor; Saba et al., 2019; Wright-Fairbanks et al., 2020; Takeshita et al., 2021). However, ISFET-based pH sensors require significant conditioning periods before deployment, suffer from biofouling, and require annual cleaning and calibration at the manufacturer as well as careful discrete sample collection at deployment and recovery to characterize and correct for sensor drift (Thompson et al., 2021). There have been few attempts to integrate *p*CO₂ sensors into gliders (Hemming et al., 2017; Hauri et al., 2018; von Oppeln-Bronikowski et al., 2021). Hemming et al. (2017) did not publish the data because of low quality. Von Oppeln-Bronikowski et al. (2021) integrated an Aanderaa CO₂ optode that measures *p*CO₂ by detecting the luminescent quenching response from a CO₂-sensitive membrane with a Slocum G2 glider but suffered from instability, thermal lag issues, variable conditioning periods (4 d to 1 month), large offsets (> 1000 μ atm), nonlinear temperature-dependent response time, and a high dependence on prior foil calibration. Hauri et al. (2018) integrated the Pro Oceanus Mini Pro CO₂ sensor with a Slocum G2. However, the Pro Oceanus Mini Pro CO₂ sensor used at the time did not withstand the pressure changes imposed by glider missions. The Franatech METS CH₄ sensor has been integrated into Alseamar SeaExplorer and Teledyne Slocum gliders and successfully used to generate concentration maps of a methane seep in a semi-quantitative way (Meurer et al., 2021).

Here we integrated modified versions of the Contros HydroC CO₂ and CH₄ sensors with a Seaglider[®] (registered trademark of the University of Washington). We discuss details of the physical and software integration, present *p*CO₂ and *p*CH₄ data from tank experiments, evaluate the quality of *p*CO₂ data collected during CO₂ Seaglider missions, and discuss highlights from missions in Resurrection Bay, Alaska.

2 Methods

2.1 CO₂ Seaglider

We integrated a modified version (Seaglider (SG) HydroC CO₂) of the CONTROS HydroC[™] CO₂ sensor (-4H-JENA engineering GmbH, Kiel, Germany) with a Seaglider M1 (Fig. 1a and b). The Seaglider M1 was specifically designed for long-endurance missions in deep waters to 1000 m depth. The HydroC CO₂ sensor was outfitted with a semi-

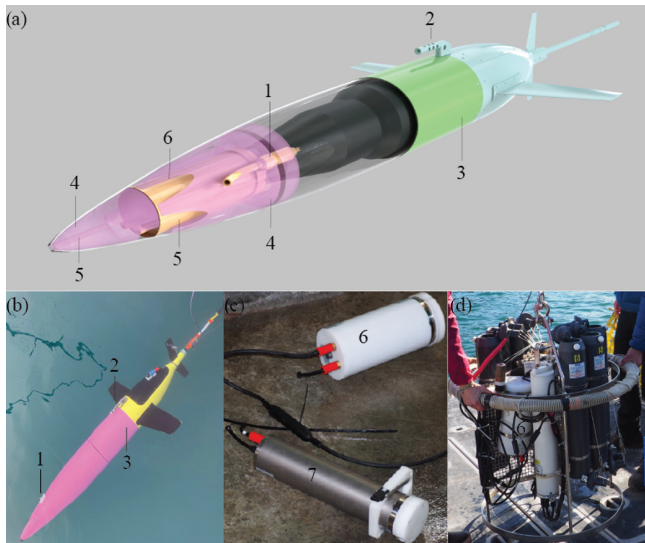


Figure 1. CO₂ Seaglider. CO₂ Seaglider (a) schematic rendering and (b) picture in Resurrection Bay, Seward, Alaska, during a checkout dive on 6 February 2023, before beginning the first winter mission collecting high-resolution *p*CO₂ data. Highlighted are the (1) Sea-Bird 5M pump, (2) conductivity and temperature sail, (3) extension, (4) syntactic foam, (5) water flow channels, and (6) SG HydroC CO₂ in a titanium housing, enabling *p*CO₂ observations down to 1000 m. (c) Picture of the new SG HydroC CO₂ in a polyoxymethylene housing (6; rated to 300 m depth) and original CONTROS HydroC™ CO₂ sensor (7). (d) Picture of the rosette setup for the profiling experiment.

permeable TOUGH membrane (Pinnau and Toy, 1996) that equilibrated dissolved CO₂ between the ambient seawater and the headspace of the sensor, where the gas concentration was determined by nondispersive infrared (NDIR) spectrometry.

Since the equilibration time (response time) of membrane-based sensors is affected by the exchange of the water mass in front of the sensor head, we installed a Sea-Bird Electronics (SBE) 5M pump next to the SG HydroC CO₂ sensor using tubing to transfer seawater from outside the glider fairing to the membrane surface (Fig. 1a). The response time was determined at the manufacturer, verified in the field, and then used to correct for hysteresis during the post-processing phase (see Sect. 2.7.2).

The form factors of the HydroC CO₂™ sensor and Seaglider were changed to achieve an internal integration of the sensor with the Seaglider. The standard high-performance HydroC CO₂™ sensor was changed from $\varnothing 89 \times 380$ to $\varnothing 136 \times 294$ mm by rearranging the gas cycle components and the control unit (Fig. 1c). This new SG HydroC CO₂ sensor is available in polyoxymethylene cladding rated to 300 m or a titanium housing rated to 1000 m to provide a choice between a coastal mission and an offshore deeper mission. Use of the titanium housing required a syntactic foam housing to compensate for the weight, whereas



Figure 2. SG HydroC CO₂ sensor mounting designs. (a) Titanium SG HydroC CO₂ (rated to 1000 m) in a custom syntactic foam coat and (b) polyoxymethylene SG HydroC CO₂ (rated to 300 m) with brackets.

the polyoxymethylene housing was integrated into the glider with simple brackets (Fig. 2). Despite these adjustments to the size of the sensor, to our knowledge, it is still the largest and heaviest sensor that has been integrated with a Seaglider to date. The forward fairing of the Seaglider was extended by 40 cm with a fiber-glass cylindrical extension to create internal wet-payload space for the sensor, pump, and cables (Fig. 1a and b). The sensor was mounted with the membrane facing aft to ensure that potential bubbles within the internal tubing of the sensor could escape the system during the downcast of the first dive. In situ comparison of the orientation of the sensor and close examination of *p*CO₂ and internal pressure data suggested that the highest data quality was achieved with this mounting design.

One of the advantages of using ocean gliders for ocean observing is the ability for real-time communication of data and commands between the pilot and the glider. To take advantage of this, modifications were needed to allow two-way communication between the Seaglider firmware and the HydroC firmware. The Seaglider firmware has a feature to allow easy integration of “logging devices”, which provides a way to build commands for the pilot on land to switch the sensor on and off and change the sampling strategy during the mission (on/off below or above certain depth) when it comes to the surface for a communication session. The Seaglider firmware can also automatically set the clock of

the sensor on request at every surfacing and send small samples of the data stream via Iridium along with the standard sensor data. This required the writing and testing of a driver file (CNF file). However, to take full advantage of the ability of the HydroC, a more advanced electronic integration was carried out using Smart Interoperable Real-time Maritime Assembly (SIRMA™, registered trademark of Cyprus Subsea Consulting and Services, C.S.C.S., Ltd.). This small programmable electronic circuit contained hardware elements to adapt the sensor power and communication requirements to those available on the host platform. It also allowed for separate storage and processing capabilities to supplement the main host processor that controls the flight, sampling, and telecommunications of the host. Most importantly here, it was programmed to relay pilot commands to the SG HydroC CO₂ for the built-in “zero” function, which isolated the internal gas circuit until there was no CO₂ present, measured the concentration signal, and assigned a zero value. Then the gas circuit was exposed to the headspace behind the diffusion membrane for in situ sampling. SIRMA was also programmed to extract raw data from the HydroC and calculate the bin average of some of the output fields, which were useful for real-time mission adaptation and confirmation of sensor operation. Three levels of output were allowed, depending on how much surfacing time could be tolerated before continuing the mission (the baud rate for Iridium is very low: on the order of 4800 bps). More detailed information can be found in the CO₂ Seaglider Standard Operating Procedure (SOP) (Irving et al., 2024).

In addition to the HydroC CO₂ sensor, the CO₂ Seaglider carried an Aanderaa 4831F optode, which is a compact optical oxygen sensor that works on the principle of luminescence quenching by oxygen with a precision of 0.1 μM and an absolute accuracy of ± 2 μM after multipoint calibration. The 4831F was equipped with a fast-response sensing foil with a well-characterized response time of 8 s. The Aanderaa optode measured absolute oxygen concentration and percentage saturation. It is the most widely used on ocean gliders and has been integrated into both Slocum gliders and Seagliders (López-García et al., 2022; Bittig et al., 2018). The Ocean-Gliders community has developed a Standard Operating Procedure (SOP) that details everything from mounting, calibration, available sensors, piloting tips, and response time correction to post-processing (López-García et al., 2022). The CO₂ Seaglider was also outfitted with an SBE CT sail and Wetlabs Ecopuck measuring chlorophyll fluorescence at 695 nm.

2.2 CH₄ Seaglider

We also integrated a modified version of the CONTROS HydroC CH₄ sensor (-4H-JENA engineering GmbH, Kiel, Germany) with the Seaglider. The manufacturer’s published uncertainty of the HydroC CH₄ sensor is 2 μatm or ± 3 %, whichever is greater. The SG HydroC CH₄ sensor had the

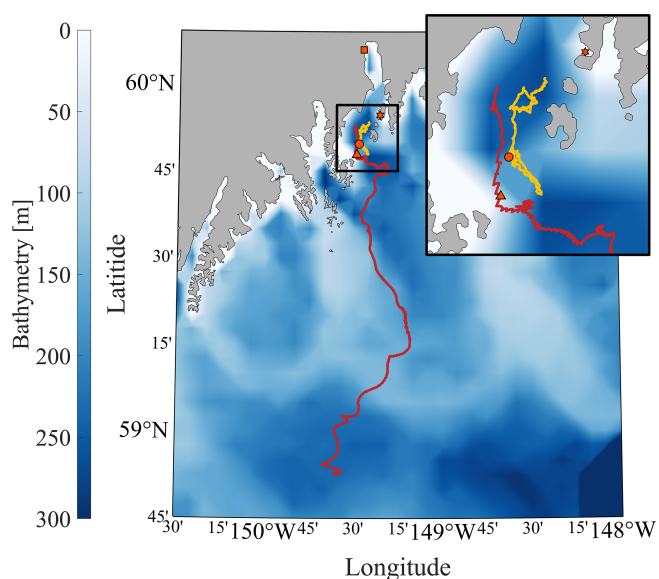


Figure 3. Map of the CO₂ Seaglider study area. The bathymetry of the Gulf of Alaska is shown in color with a zoomed-in section of the head of Resurrection Bay (outlined black square and inset map). Tracks of the CO₂ Seaglider from the 4–7 May 2022 and 8–21 February 2023 missions are shown in yellow and red, respectively. Orange markers outlined in black show the location of the Alutiiq Pride Marine Institute (square), the National Oceanic and Atmospheric Administration’s Gulf of Alaska Ocean Acidification mooring (star), the 7 May CTD cast (circle), and the last location where *p*CO₂ data were collected during the February 2023 mission (triangle).

same form factor as the SG HydroC CO₂ sensor. However, it was 0.5 kg heavier due to its tunable diode laser absorption spectroscopy (TDLAS) component, so the SG HydroC CH₄ had to be integrated with changes to the glider’s ballast.

2.3 Spring and winter CO₂ Seaglider missions

Both versions of the CO₂ Seaglider (rated to 300 m versus 1000 m) were tested in separate missions (Fig. 3) in spring (53 dives from 4–7 May 2022; Fig. 4) and winter (310 dives from 8–21 February 2023; Fig. 5). The 300 m version with integrated polyoxymethylene housing was tested during the 4–7 May 2022 mission. The glider followed along a transect within Resurrection Bay. Conductivity–temperature–depth (CTD) casts near the glider path allowed for in-depth evaluation of the data quality. The 1000 m depth-rated CO₂ Seaglider with integrated titanium housing was tested in February 2023. Estimated energy consumption during the CO₂ Seaglider missions was 19 out of 135 Ah and 75 out of 120 Ah for the 24 V which powered the SG HydroC CO₂ sensor battery for the spring and winter missions, respectively. Before the February mission, the onboard modem was replaced with a newer model, with different input voltage requirements, which were probably not met as the

mission evolved. As a result, the glider could not communicate and was lost. While this was an unfortunate mistake, the loss of the glider had nothing to do with the HydroC CO₂ integration.

2.4 Tank experiments

Shortly before the May 2022 glider mission, the glider was kept in a flow-through tank at the Alutiiq Pride Marine Institute for roughly 12 h for cross-calibration purposes. The flow-through tank was fed with water from about 75 m depth and 91 m from the laboratory into Resurrection Bay, near a freshwater source. During the tank experiment, SG HydroC CO₂T-0718-001 (Fig. 6b, blue line) was integrated into the Seaglider, and the SG HydroC CO₂T-0422-001 (Fig. 6b, black line) and SG HydroC CH₄ (Fig. 6c) sensors were secured next to the Seaglider. The water was kept in motion with a circulation pump. Triplicate discrete water samples for dissolved inorganic carbon, pH, and *p*CH₄ analysis were taken every 4 h (Table 1).

2.5 Rosette package

One of the SG HydroC CO₂ sensors (CO₂T-0422-001) was installed on an SBE-55 frame ECO water sampler with six 4 L sample bottles (Sea-Bird Scientific) during the May 2022 trials (Tables 2 and 3; Figs. 7 and S1). The SBE-55 and SG HydroC CO₂ were powered by an SBE-33 carousel deck unit. The SG HydroC CO₂ interfered with the communication stream and thereby prevented real-time data acquisition and control of the SBE-55, but data were internally logged. The depth of the rosette package was monitored directly on the winch, and the timing of firing of the sample bottles, after an approximate 15 min hovering period (to allow for equilibration), was programmed in advance based on time intervals. On 3 May (Table 2; Fig. 7) only samples from the upper 20 m of the water column were usable due to issues with manually measuring the depths and the sample collection. On 7 May (Table 3; Fig. S1) two bottles that were intended to be fired while the rosette was stationary at depth, were instead fired while the rosette was in motion.

2.6 Discrete water samples

2.6.1 Inorganic carbon chemistry

Discrete seawater samples were collected for sensor validation in two different cases in May of 2022. Firstly, samples were taken alongside two SG HydroC CO₂ sensors during a tank experiment at the Alutiiq Pride Marine Institute (Fig. 6b; Table 1) (Fig. 1d). Secondly, samples were taken from bottles during a CTD cast within 1 km and 4 h of the HydroC measuring *p*CO₂ on the glider while conducting dives (Sect. 3.2).

Inorganic carbon sampling in the Gulf of Alaska's glaciated coastal regions requires methodological variations from open-ocean best practices to ensure that suspended min-

eral particles do not compromise the instrumentation and/or bias measurements between sample collection and analysis (Sejr et al., 2011). Given this, the discrete seawater samples were filtered (replaceable 0.45 μm filter in a 47 mm polycarbonate in-line filter) with a peristaltic pump straight from the Niskin bottles (see Bockmon and Dickson, 2014, for detailed methods) or tank into pre-cleaned 500 mL borosilicate bottles and poisoned with 200 μL mercuric chloride (HgCl₂) (Dickson et al., 2007). Samples were transported and stored at room temperature before analysis. Samples were opened immediately (< 10 min) before concurrent analyses of pH and dissolved inorganic carbon (DIC) to limit gas exchange with ambient lab conditions. Samples were analyzed for DIC using an Apollo SciTech, LLC dissolved inorganic carbon analyzer model AS-C6. All species of dissolved inorganic carbon in a sample were converted to CO₂ by the addition of a strong acid. The CO₂ gas was then purged from the sample through a drying system. The concentration of CO₂ gas was measured using a nondispersive infrared gas analyzer, the LI-7000 CO₂ / H₂O analyzer. This method required Certified Reference Material (CRM; batch 198 from Andrew Dickson's Certified Reference Materials Laboratory) to create a three-point calibration line. The calibration line was used to quantify the total amount of CO₂ in the sample as the integrated area under the concentration–time curve. Apollo SciTech recommendations to improve analytical accuracy were followed and included bubbling of CO₂ off the acid daily; allowing the analyzer to warm up for at least 2 h before measurements begin; measuring a set of standards at the beginning and end of each day and every nine samples; using ultrahigh-purity (UHP) N₂ gas; and filtering the N₂ gas with a PTFE filter, CO₂ scrubber (Ascarite II), and H₂O scrubber (Mg(ClO₄)₂).

Samples were analyzed spectrophotometrically for pH with a CONTROS HydroFIA pH (Aßmann et al., 2011) operating in discrete measurement mode using unpurified metacresol purple (mCP) as the indicator dye (Clayton and Byrne, 1993). Sample temperature was stabilized at 25.00 ± 0.01 °C during measurements using Peltier elements, and five repetitive measurements were taken for each sample. At the beginning of each day, the HydroFIA pH underwent a conditioning period using seawater with similar properties until values stabilized. CRMs (known TA and DIC concentration; batch 198 from Andrew Dickson's Certified Reference Materials Laboratory) were measured at the beginning and end of the day, as well as every nine samples.

All data processing and analyses were done using an in-house MATLAB routine. In situ pH and *p*CO₂^{disc} were calculated from input pair pH_{lab} and DIC using CO₂SYSv3 (Sharp et al., 2023) with dissociation constants for carbonic acid of Sulpis et al. (2020), bisulfate of Dickson (1990), hydrofluoric acid of Perez and Fraga (1987), and the boron-to-chlorinity ratio of Lee et al. (2010). pH_{lab} is defined as the pH measured on the total scale at measurement temperature and 1 atm of pressure (0 dbar applied pressure) us-

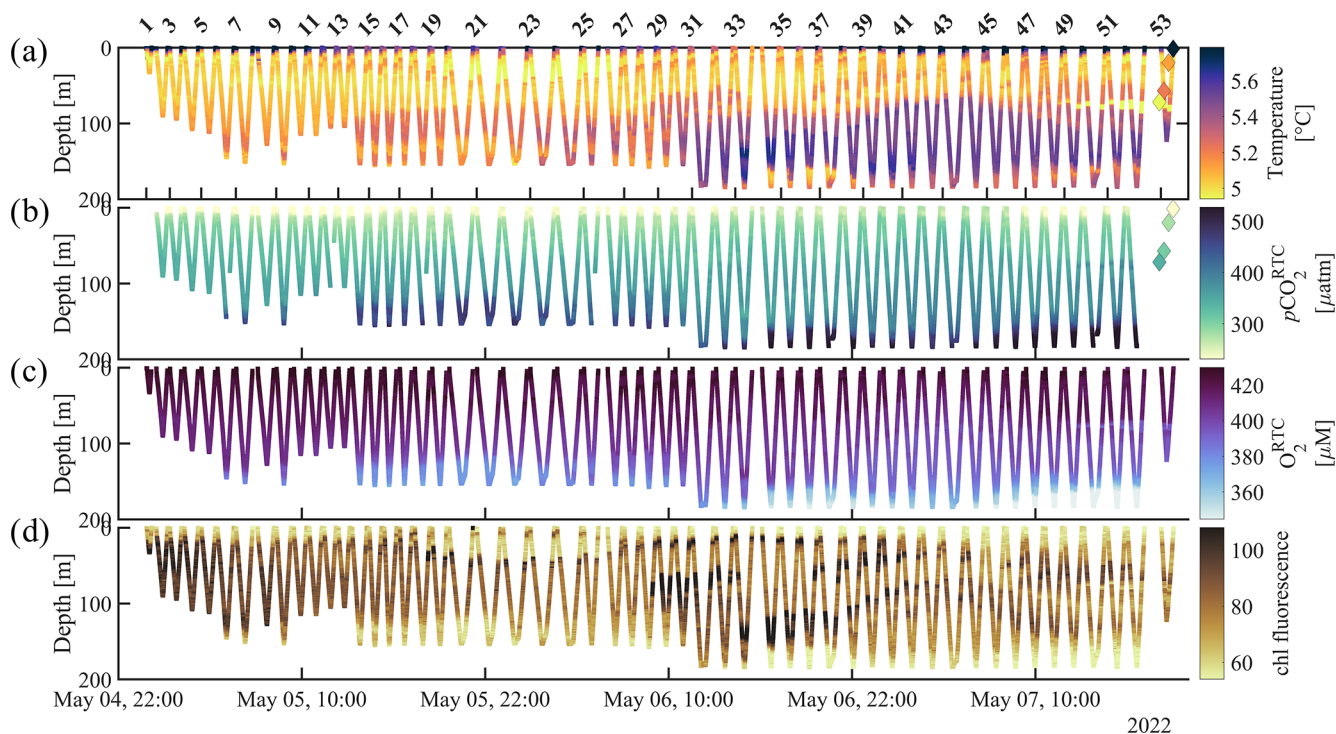


Figure 4. CO₂ Seaglider data from the 4–7 May 2022 mission in Resurrection Bay, Seward, Alaska. Depth profiles of (a) temperature (°C), (b) response-time-corrected $p\text{CO}_2$ ($p\text{CO}_2^{\text{RTC}}$, μatm), (c) response-time-corrected O₂ (O_2^{RTC} , μM), and (d) raw chlorophyll fluorescence. The diamonds show discrete values that were taken during a CTD cast (Table 3).

Table 1. Tank experiment. Evaluation of SG HydroC CO₂ and SG HydroC CH₄ sensors compared to reference discrete $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$. Units of $p\text{CO}_2$ and $p\text{CH}_4$ are micro-atmospheres (μatm) except when shown as a percent difference in parentheses (Eq. 1). Columns with subscripts sn422 and sn0718 indicate data from sensors HydroC CO₂T-0422-001 and HydroC CO₂T-0718-001, respectively. The superscript RTC indicates response-time-corrected values following Dølvén et al. (2022). $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$ values are the average of triplicate bottles and are shown in Fig. 6.

Triplicate date and time (UTC)	$p\text{CO}_2^{\text{disc}} \pm u_c$ (μatm)	$p\text{CO}_2^{\text{RTC}}_{\text{sn422}} - p\text{CO}_2^{\text{disc}}$	$p\text{CO}_2^{\text{RTC}}_{\text{sn0718}} - p\text{CO}_2^{\text{disc}}$	$p\text{CH}_4^{\text{disc}} \pm u$ (μatm)	$p\text{CH}_4^{\text{RTC}} - p\text{CH}_4^{\text{disc}}$
5 Feb 2022, 03:25	298.7 ± 10.2	-0.9 (−0.3 %)	–	–	–
5 Feb 2022, 07:32	227.1 ± 7.8	4.3 (1.9 %)	2.4 (1.1 %)	–	–
5 Feb 2022, 11:27	223.3 ± 7.7	0.7 (0.3 %)	-2.6 (−1.2 %)	–	–
5 Feb 2022, 15:30	227.8 ± 7.9	-1.1 (−0.5 %)	-3.3 (−1.4 %)	–	–
5 Feb 2022, 00:11	–	–	–	25.4 ± 2.1	4.0 (15.8 %)
5 Feb 2022, 12:06	–	–	–	7.3 ± 1.3	0.5 (6.6 %)

ing spectrophotometric methods, with instrument specifications described above. Sulpis et al. (2020) found that the carbonic acid dissociation constants of Lueker et al. (2000) may underestimate $p\text{CO}_2$ in cold regions (below $\sim 8^\circ\text{C}$) and therefore overestimate pH and carbonate ion concentrations (CO_3^{2-}). Differences between discrete $p\text{CO}_2$ calculated with the carbonic acid dissociation constants by Lueker et al. (2000) (the standard in synthesis data products; e.g., Jiang et al., 2021; Lauvset et al., 2022; Metzl et al., 2024) and the HydroC $p\text{CO}_2$ from the tank experiment were found to be

on average $4.6 \mu\text{atm}$ (1.6 %) and $4.2 \mu\text{atm}$ (0.7 %) greater for SN0422 and SN0718, respectively, when compared with discrete $p\text{CO}_2$ based on carbonic acid dissociation constants by Sulpis et al. (2020).

Discrete $p\text{CO}_2$ uncertainty (u_c) was calculated as the combined standard uncertainty from errors.m (Orr et al., 2018) that propagates input uncertainties plus errors in the dissociation constants. Input uncertainties for pH_{lab} and DIC were the standard uncertainties, defined as the square root of the sum of the squared random uncertainty component plus the

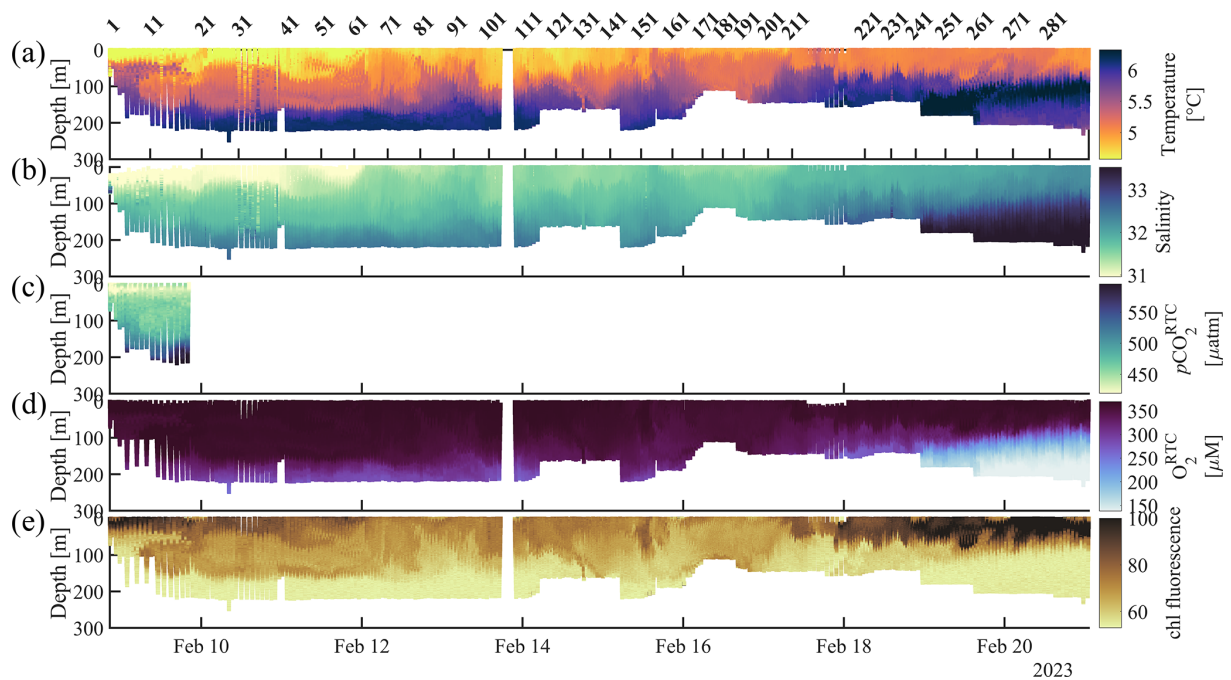


Figure 5. CO₂ Seaglider data collected during the 8–21 February 2023 winter mission. Shown are (a) temperature (°C), (b) salinity, (c) response-time-corrected $p\text{CO}_2$ ($p\text{CO}_2^{\text{RTC}}$, μatm), (d) response-time-corrected oxygen (O_2^{RTC} , μM), and (e) raw chlorophyll fluorescence (chl fluorescence) as time and dive number vs. pressure.

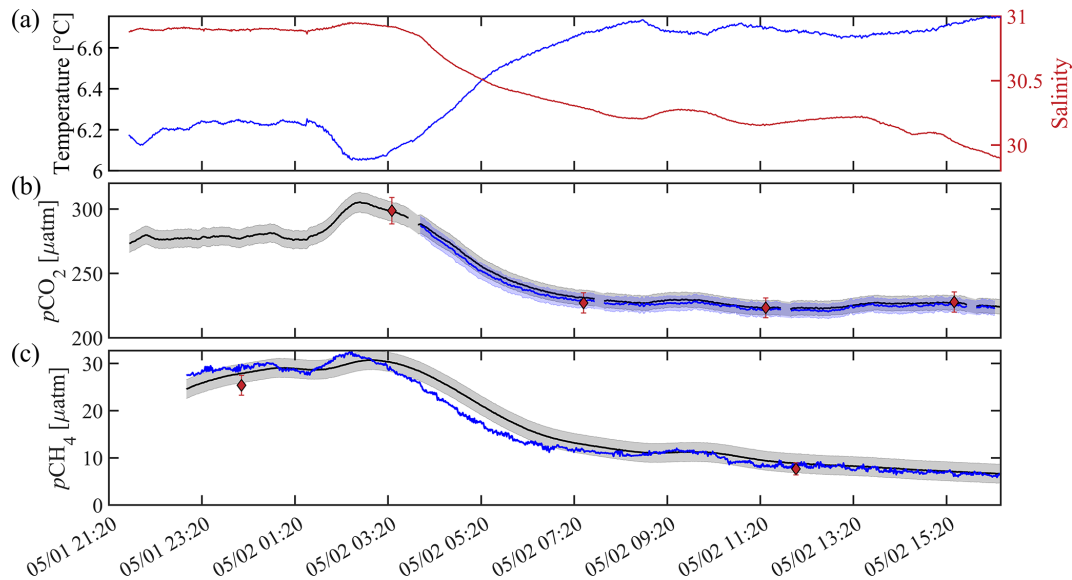


Figure 6. Sensor validation during a tank experiment at the Alutiq Pride Marine Institute on 1–2 May 2022. (a) Temperature (blue line) and salinity (red line) from a recently calibrated Sea-Bird Scientific SBE37. (b) Black (blue) lines show $p\text{CO}_2$ in micro-atmospheres (μatm) from HydroC CO2T-0422-001 (HydroC CO2T-0718-001), with the shaded gray (blue) areas showing a relative uncertainty of 2.5 % (weather quality goal; Newton et al., 2015). Black circles with red filling show discrete $p\text{CO}_2^{\text{disc}}$, with error bars showing the combined standard uncertainty from errors.m (Orr et al., 2018). HydroC $p\text{CO}_2$ data are shown at 1 min resolution with a 2 min moving median filter applied and have not been corrected for response time, but differences were negligible ($< 0.1 \mu\text{atm}$). (c) The black line shows $p\text{CH}_4$ in micro-atmospheres (μatm) from HydroC CH₄T-0422-001, with the shaded gray bar showing an uncertainty of 2 μatm . The blue line is the response-time-corrected signal with a response time of 43 min following Dølven et al. (2022). HydroC $p\text{CH}_4$ data are shown at 1 min resolution with a 2 min moving median filter applied to the raw data and a 10 min moving median filter applied to the RTC data. Black diamonds with red filling show discrete $p\text{CH}_4^{\text{disc}}$, and all discrete values of $p\text{CO}_2^{\text{disc}}$ and $p\text{CH}_4^{\text{disc}}$ are the average of triplicate bottles.

Table 2. Profiling experiment. Evaluation of the SG HydroC CO₂ sensor compared to reference discrete $p\text{CO}_2^{\text{disc}}$. Units of $p\text{CO}_2$ are micro-atmospheres (μatm) except when shown as a percent difference in parentheses (Eq. 1). $p\text{CO}_2$ with the subscript “Rosette” indicates data from the HydroC sensor mounted on the rosette (HydroC CO2T-0422-001). The superscript RTC indicates response-time-corrected values following Dølvén et al. (2022).

Discrete date and time (UTC)	Discrete depth (m)	$p\text{CO}_2^{\text{disc}} \pm uc$ (μatm)	$p\text{CO}_{2,\text{Rosette}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$
5 Mar 2022, 21:21	2.5	214.5 ± 7.5	5.4 (2.5 %)
5 Mar 2022, 21:39	19.9	246.8 ± 8.5	1.6 (0.6 %)
5 Mar 2022, 22:33	9.6	244.4 ± 8.5	-3.3 (-1.4 %)
5 Mar 2022, 22:34	9.7	234.7 ± 8.1	8.2 (3.5 %)

Table 3. Seaglider HydroC evaluation with a nearby cast. Evaluation of Seaglider-integrated and rosette-mounted SG HydroC CO₂ sensors compared to $p\text{CO}_2^{\text{disc}}$ collected from a nearby cast. Units of $p\text{CO}_2$ are micro-atmospheres (μatm) except when shown as a percent difference in parentheses (Eq. 1); differences of $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ were calculated using the average (upcast and downcast combined) 1 m binned data. The superscript RTC indicates response-time-corrected values following Dølvén et al. (2022), and the subscripts “Rosette” and “Seaglider” indicate the SG HydroC CO₂ sensor mounted on the rosette (SG HydroC CO2T-0422-001) and integrated into the Seaglider (SG HydroC CO2T-0718-001), respectively. The time delay (HH:MM) and spatial distance (km) columns represent the distance of $p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}}$ measured at the discrete depth and the discrete date and time. The asterisk (*) indicates the comparison with $p\text{CO}_{2,\text{Rosette}}^{\text{RTC}}$ taken as nearest in time before sensor zeroing (Fig. S1).

Discrete date and time (UTC)	Discrete depth (m)	$p\text{CO}_2^{\text{disc}} \pm uc$ (μatm)	$p\text{CO}_{2,\text{Rosette}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$	Delay (HH:MM)	Distance (km)	$p\text{CO}_{2,\text{Seaglider}}^{\text{RTC}} - p\text{CO}_2^{\text{disc}}$
5 Jul 2022, 18:06	71.8	349.7 ± 7.8	-5.7 (-1.6 %)	02:47	0.4	10.2 (2.9 %)
5 Jul 2022, 18:24	57.1	313.8 ± 6.7	12.1 (3.9 %)	03:05	0.6	8.3 (2.7 %)
5 Jul 2022, 18:42	19.8	285.3 ± 6.1	0.8 (0.3 %)	03:23	0.8	8.6 (3.0 %)
5 Jul 2022, 19:00	1.6	233.4 ± 5.0	-2.3 (-1.0 %)*	03:41	0.9	12.0 (5.1 %)

squared systematic uncertainty components. For pH_{lab} the random uncertainty was the sample precision, or standard deviation of the measurements. For DIC, the random uncertainty was the propagated error calculated with the first-order Taylor series expansion (Eq. 1; Orr et al., 2018) and assuming the correlation term was zero for the conversion of molarity ($\mu\text{moles L}^{-1}$) to molality ($\mu\text{moles kg}^{-1}$). Systematic uncertainty components were the uncertainty in the CRM used for instrument offset and drift correction, as well as the published instrument accuracy, or, if available, the daily instrument accuracy as defined below. Daily instrument accuracy was defined as the maximum difference between the known CRM concentration and the measured CRM concentration after data were corrected for instrument drift and offset of all available CRMs not used in the instrument drift and offset calculation. CRM pH_{lab} “known” values were calculated using CO2SYSv3 (Sharp et al., 2023) with inputs TA and DIC. Nutrient concentrations (SiO_4^{-2} , PO_4^{-3}) were assumed to be negligible in the CO2SYS calculations (e.g., DeGrandpre et al., 2019; Vergara-Jara et al., 2019; Islam et al., 2017).

2.6.2 Methane

Two sets of triplicate $p\text{CH}_4$ discrete water samples were collected during the tank experiment for an initial evaluation of the SG HydroC CH₄ sensor (Table 1, Fig. 6c). Seawater was filtered from the tank into 250 mL vials. The vials were closed with a rubber stopper, topped with an aluminum cap, and closed with a crimp immediately. A dry and clean syringe was flushed with 10 mL of N₂ gas twice. The third fill was kept, and the syringe valve was closed. N₂ was then injected into the headspace while simultaneously pulling 10 mL of seawater out of the vial using a second syringe. 50 μL mercuric chloride (HgCl_2) was added to the vial, which was then shaken for about a minute and put into a fridge at 4 °C for over 12 h to equilibrate the headspace. The samples were then sent to the Kessler analytical laboratory at the University of Rochester for methane concentration analysis following previously published procedures (Leonte et al., 2020).

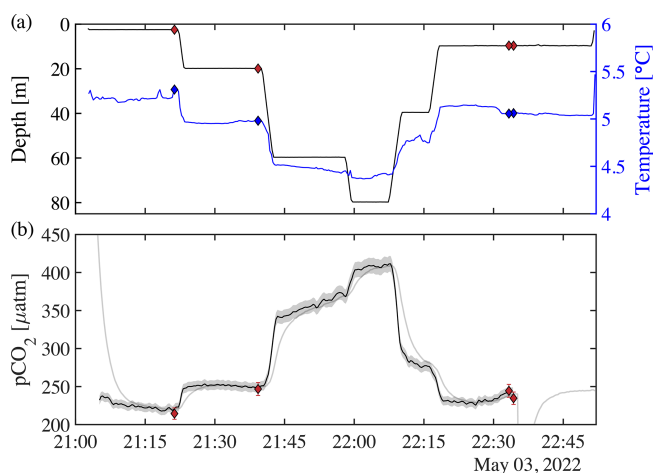


Figure 7. Profiling experiments from 3 May 2022 with the HydroC CO2T-0422-001 sensor mounted on the rosette. **(a)** Pressure vs. time on the left (black) axis, with diamonds showing rosette CTD values of pressure (filled red diamond), and temperature vs. time on the right (blue) axis as well as temperature (filled blue diamond) at the time of the bottle fire. **(b)** $p\text{CO}_2$ measured by the rosette-mounted SG HydroC CO_2 sensor as a raw (gray line) and response-time-corrected signal (thick black line; $p\text{CO}_{2,\text{Rosette}}^{\text{RTC}}$ in Table 2), with shaded relative uncertainty of 2.5% (weather goal; Newton et al., 2015). $p\text{CO}_2^{\text{disc}}$ shown as red diamonds, with vertical red error bars showing combined standard uncertainty (Orr et al., 2018). Table 2 shows differences between discrete $p\text{CO}_2^{\text{disc}}$ and $p\text{CO}_{2,\text{Rosette}}^{\text{RTC}}$. The SG HydroC CO_2 sensor started a zeroing interval at 22:35 on 3 May 2022, so $p\text{CO}_{2,\text{Rosette}}^{\text{RTC}}$ is not shown after that time but signal recovery can be seen in the uncorrected signal (gray line).

2.7 Data post-processing

2.7.1 $p\text{CO}_2$ post-processing

SG HydroC CO_2 data were post-processed using Jupyter Notebook scripts developed by -4H-JENA engineering GmbH at the original resolution (2 s). SG HydroC CO_2 (SG HydroC CO2T-0422-001) data from the tank experiment (Table 1; Fig. 6) and rosette-mounted CTD casts (Tables 2 and 3; Figs. 7 and S1) were post-processed to correct for baseline drift (change in the zero signal reference) and span drift (changes in the sensor's concentration dependent characteristics) using pre- and post-calibration coefficients interpolated over the deployment (Fietzek et al., 2014). For the May 2022 Seaglider-integrated SG HydroC CO_2 sensor (SG HydroC CO2T-0718-001; Table 3; Figs. 4 and 8), data were post-processed with pre-calibration coefficients only (no span drift correction) because the sensor was damaged during the return shipment for post-calibration. Differences between sensors remained low despite the difference in processing, with a mean difference during the tank experiment of $2.1 \pm 1.0 \mu\text{atm}$ (0.9%) and median difference of

$2.0 \pm 1.0 \mu\text{atm}$ (0.9%) (Table 1; Fig. 6b). The $p\text{CO}_2$ data from February 2023 were collected with a sensor that was factory-calibrated 2 weeks prior to deployment (SG HydroC CO2T-0422-001) but were not post-processed because a required parameter (p_{NDIR}) was not relayed in real time and the glider was lost. Lack of post-calibration most likely had no negative effect on the quality of data since the HydroC only collected data for ~ 4 d during the spring mission and ~ 2 d during the winter mission.

HydroC $p\text{CO}_2$ and $p\text{CO}_2^{\text{RTC}}$ data at the original resolution (2 s) and RTC resolution (8 s) were linearly interpolated onto the Seaglider timestamp, and 1 m binned data were calculated by first averaging 1 m (± 0.5 m) upcast and downcast data independently, linearly interpolating over gaps, and then averaging the interpolated 1 m binned upcasts and downcasts together.

2.7.2 Response time correction

The ability to determine the in situ response time (τ_{63} of the HydroC, which took into account membrane characteristics and the rate of water exchange over the membrane, i.e., pump characteristics) of the sensor made correction for hysteresis through data post-processing possible. This is critical for a sensor operating on profiling platforms, especially in the Gulf of Alaska, where strong environmental gradients were encountered. Fiedler et al. (2013) used a CONTROS HydroCTM CO_2 with a silicone polydimethylsiloxane (PDMS) membrane and reported a linear response time dependency on water temperature on the order of 1 s per 1°C . For this study, the SG HydroC CO_2 sensors were deployed with the new robust TOUGH membrane, which had Teflon AF2400 as the active separation layer with a low temperature dependence on the permeability coefficient (Pinnau and Toy, 1996). Response times determined during calibration at -4H-JENA were used for response time correction (RTC) and found to be 106 s for the HydroC mounted on the rosette in May 2022 and 108 s when it was integrated into the Seaglider in February 2023 (HydroC CO2T-0422-001). The response time of the HydroC integrated into the Seaglider in May 2022 (HydroC CO2T-0718-001) was 109 s. Since field verification of the response time was recommended to ensure the highest-quality post-processed data product (because τ_{63} can be affected by the speed of water exchange across the membrane due to pump speed, tube length, etc.), we verified the sensor response time at deployment. After the glider was stationary for approximately 15 min, a zeroing interval was performed with the HydroC CO_2 sensor. The response time was determined by reviewing the time it took for the signal to recover to the ambient concentration. Our in situ response time tests were suggested to be within 5 s of the response time found during calibration (not shown). Before RTC was applied, HydroC CO_2 data were smoothed using a quadratic regression (MATLAB's smoothdata.m function with the loess method) over a 2 min window. This was done to eliminate

erroneous spikes in the RTC signal while retaining the original 2 s resolution of the $p\text{CO}_2$ data. The RTC resolution of 8 s was determined with the L-curve analysis included in the publicly available code from Dølvén et al. (2022). The Dølvén et al. (2022) RTC method was used because it produced more realistic profiles than an RTC method (Miloševich et al., 2004; not shown) previously used for HydroC CO_2 correction from a profiling float (Fielder et al., 2013). In addition, Dølvén et al. (2022) developed their algorithm with equilibrium-based sensors in mind, and it was proven with a sensor with a long response time (HydroC CH_4 $\tau_{63} \cong 23$ min).

2.7.3 $p\text{CH}_4$ post-processing

SG HydroC CH_4 data were response-time-corrected using a τ_{63} of 43 min (Dølvén et al., 2022; Fig. 6c, blue line). Before RTC was applied, HydroC $p\text{CH}_4$ data were smoothed using a quadratic regression (MATLAB's `smoothdata.m` function with the Loess method) over a 2 min window to avoid erroneous spikes in the RTC data while retaining the original 2 s resolution of the $p\text{CH}_4$ data. The RTC resolution of 30 s was determined with the L-curve analysis included in the publicly available code from Dølvén et al. (2022). Discrete $p\text{CH}_4$ samples were collected during the tank experiment (Table 1; Fig. 6c, red diamonds) and analyzed at the Kessler analytical laboratory at the University of Rochester for methane concentration analysis following previously published procedures (Leonte et al., 2020). Discrete $p\text{CH}_4$ sample values were converted from the concentration of dissolved gas in water (mol L^{-1}) to partial pressure ($p\text{CH}_4^{\text{disc}}$, μatm) using the solubility coefficient following Sarmiento and Gruber (2006). $p\text{CH}_4^{\text{disc}}$ uncertainty (u ; Table 1; Fig. 6c, red error bars) was calculated as the square root of the sum of the squared (1) mean of the standard deviations from each sample as returned from the lab and (2) the standard deviation of the triplicates.

3 Results

3.1 Glider flight

The CO_2 Seaglider was able to “fly” properly, allowing the desired undisturbed flow, despite the large payload and major changes to the vehicle fairing. Example flight profiles with the polyoxymethylene and titanium integrated sensors are shown in Figs. 9 and 10, respectively. Pitch and vertical velocity are in the stable range and roughly symmetric between downcast and upcast, indicating a nearly balanced glider. Heading varies around the targeted value as the roll adjusted to heading errors. It should be noted that this level of variability is typical of standard Seagliders. Operating Seagliders in shallow water (< 200 m) is risky because of the likelihood of meeting depth-averaged currents of the same order of magnitude as the vehicle speed. A typical sin-

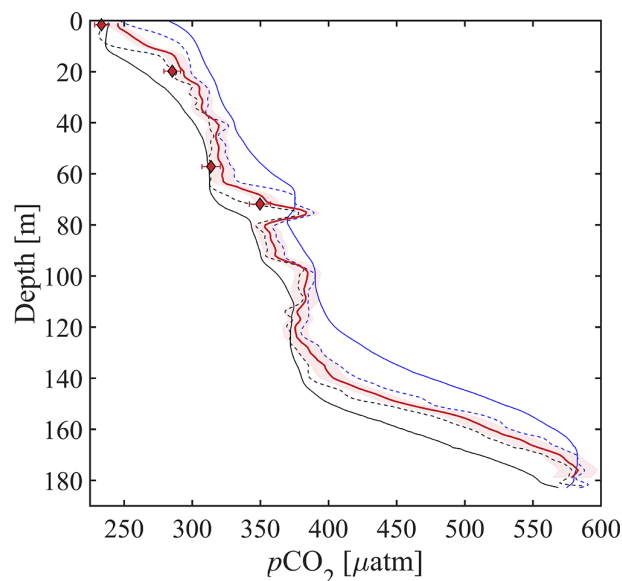


Figure 8. CO_2 Seaglider data from a single dive during the 4–7 May 2022 mission in Resurrection Bay, Seward, Alaska. Depth profile of $p\text{CO}_2$ in micro-atmospheres (μatm) showing the original-resolution smoothed $p\text{CO}_2$ (downcast: solid black, upcast: solid blue), RTC $p\text{CO}_2$ following Dølvén et al. (2022) (dashed black line: downcast, dashed blue line: upcast), and 1 m binned RTC profile (thick red line) with red shading showing the relative uncertainty of 2.5%. Discrete $p\text{CO}_2$ ($p\text{CO}_2^{\text{disc}}$) is shown as red diamonds, with horizontal red error bars showing combined standard uncertainty (Orr et al., 2018). Differences between $p\text{CO}_2^{\text{disc}}$ and $p\text{CO}_2^{\text{RTC}}$ are shown in Table 3.

gle dive cycle of downcast and upcast shows that the sensor data are free of noise that could be expected if there were recirculated water from the glider meeting the sensors. The expected endurance of the CO_2 Seaglider is around 18 and 15 d for the CH_4 Seaglider with constant sampling at full depth.

3.2 CO_2 Seaglider data evaluation

The quality of the CO_2 Seaglider data was thoroughly tested with discrete measurements during a tank experiment, nearby CTD cast, and glider missions.

3.2.1 Tank experiment

Discrete water samples show good agreement with the SG HydroC CO_2 sensors (Fig. 6b; Table 1). The values of discrete water samples represent the average of triplicate samples (Fig. 6c, red diamonds). Differences between the SG HydroC CO_2 sensors remained low, with a mean difference during the tank experiment of $2.1 \pm 1.0 \mu\text{atm}$ (0.9%) and median difference of $2.0 \mu\text{atm}$ (0.9%; Table 1). Percent differences (Eq. 1) between the SG HydroC CO_2 sensors and discrete water samples collected in the tank were between

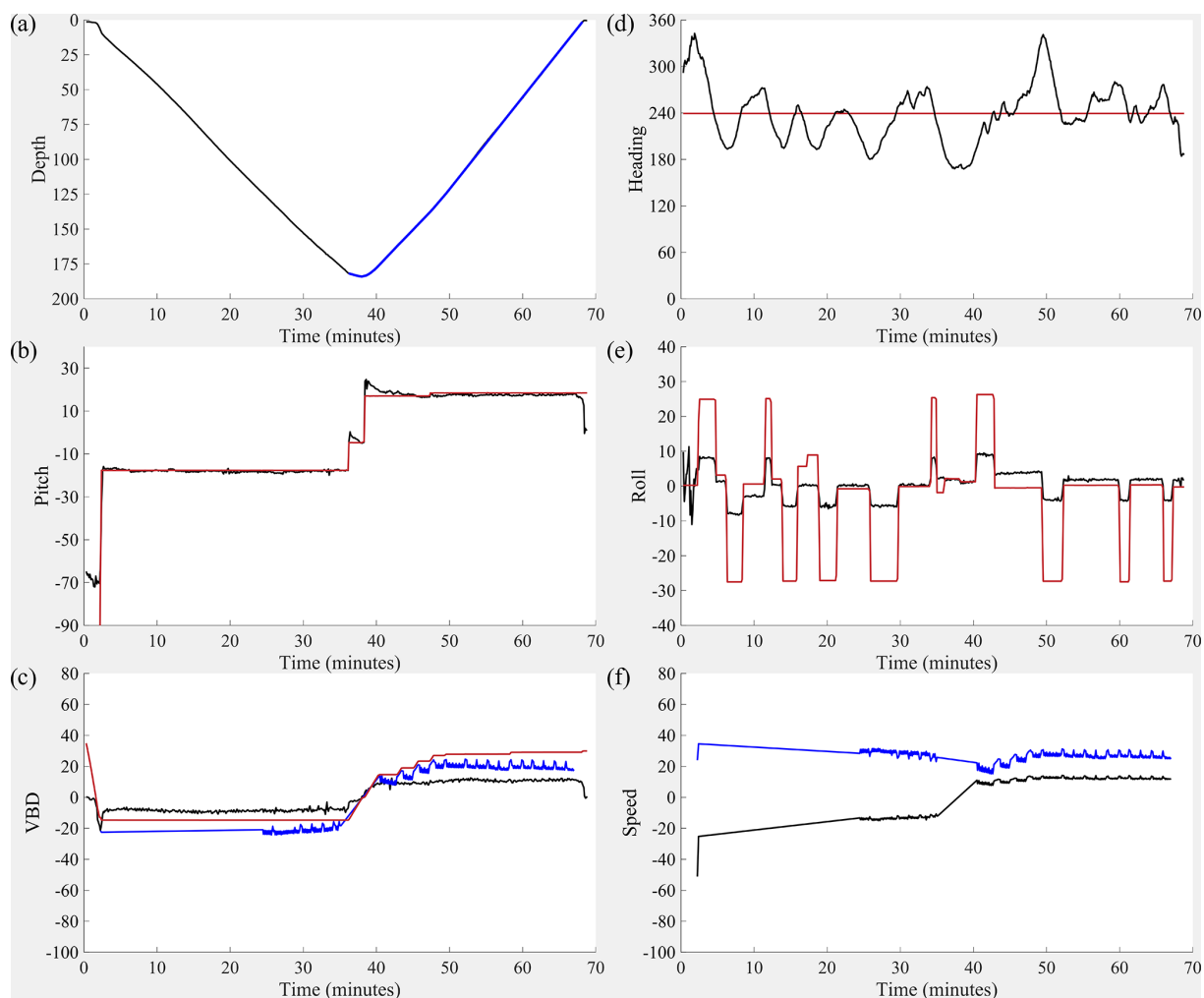


Figure 9. Dive details for the 300 m rated CO₂ Seaglider (dive 51). **(a)** Depth (black line; meters), **(b)** pitch (black line; degrees) with pitch control (red line; mm of battery shift), **(c)** change in displacement of variable buoyancy drive (VBD) (red line; units of 10 cc), vertical velocity from pressure measurements (black line; cm s⁻¹), buoyancy (blue line; units of 10 g), **(d)** heading (desired: red line, measured: black line; degrees), **(e)** roll (battery roll position: red line, glider-measured roll: black line; degrees), and **(f)** vertical speed (calculated from buoyancy and pitch; black line; cm s⁻¹) and horizontal speed (calculated from buoyancy and pitch; blue line; cm s⁻¹).

−1.4 % and 1.9 % (Table 1; Fig. 6).

$$\% \text{ difference} = \frac{p\text{CO}_2^{\text{HydroC}} - p\text{CO}_2^{\text{disc}}}{p\text{CO}_2^{\text{disc}}} \cdot 100\% \quad (1)$$

3.2.2 Profiling experiment

Rosette-based profiles with the SG HydroC CO₂ sensor in combination with discrete water samples were used to test and evaluate the response time correction algorithm by Dølvén et al. (2022). The rosette was lowered into the water and kept at different depths for about 20 min at a time (Figs. 7a and S1a). Sample bottles were programmed to collect seawater toward the end of each hovering period. *p*CO₂ measured with the HydroC ranged from 218 μatm at the surface to 411 μatm at 80 m depth on 3 May (Fig. 7b)

and 231 μatm at the surface to 382 μatm at 77 m depth on 7 May (Fig. S1). Differences between the rosette-mounted SG HydroC CO₂ sensor and discrete samples ranged from −3.3 μatm (−1.4 %) to 8.2 μatm (3.5 %), with the lowest percent difference of 0.6 % (Table 2) on 3 May, and from −5.7 μatm (−1.6 %) to 12.1 μatm (3.9 %), with the lowest percent difference of 0.3 % (Table 3) on 7 May.

3.2.3 Data evaluation during CO₂ Seaglider mission

The quality of the *p*CO₂ data from the CO₂ Seaglider was further evaluated during a 4–7 May sea trial mission in spring 2022 in Resurrection Bay, Alaska (Fig. 3).

Discrete water samples were taken in proximity (1 km and within 4 h) to the downcast of dive 51 (Table 3; Figs. 4a and 8). The response-time-corrected CO₂ Seaglider data com-

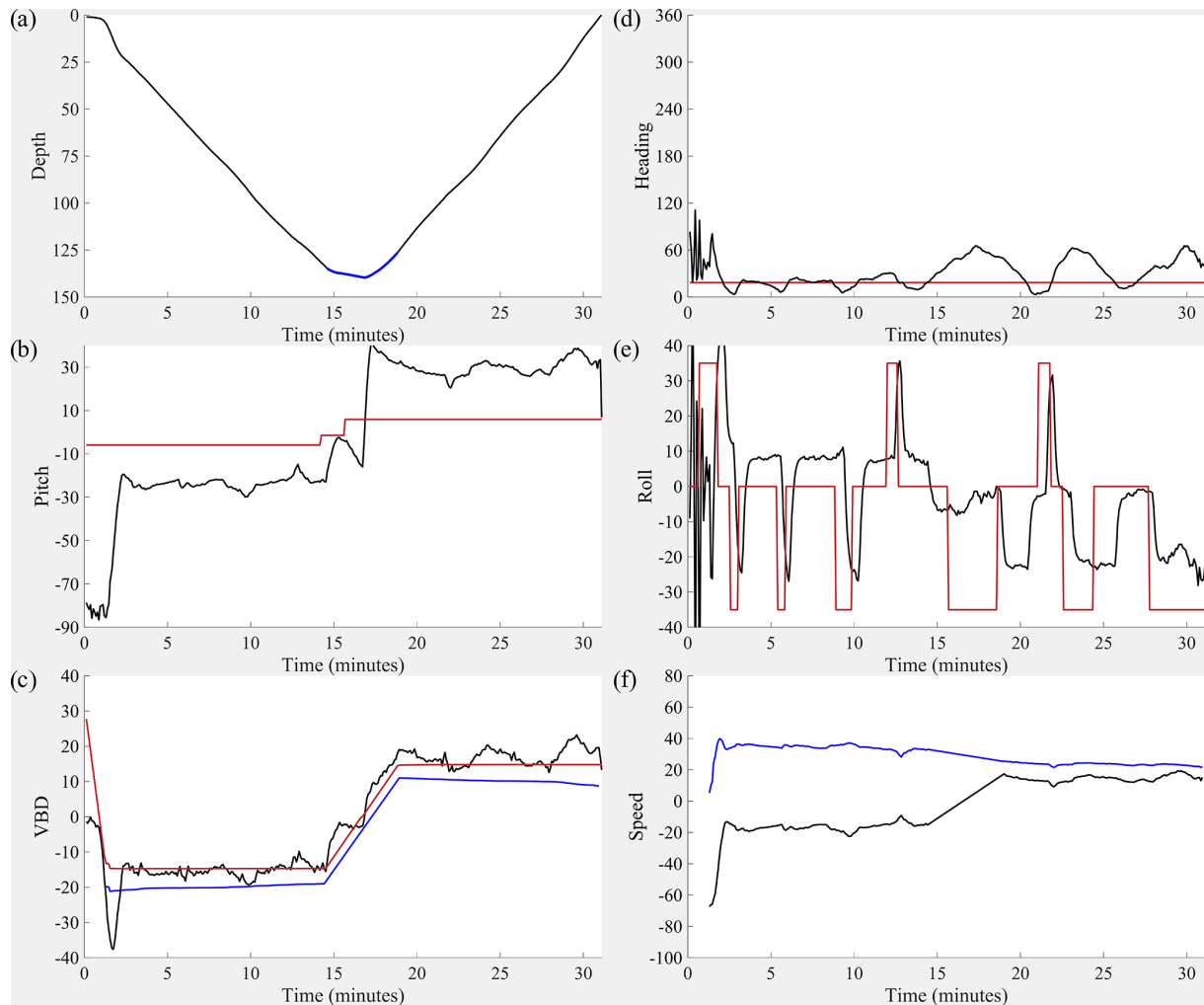


Figure 10. Dive details for the 1000 m rated CO₂ Seaglider (dive 203). **(a)** Depth (black line; meters), **(b)** pitch (black line; degrees) with pitch control (red line; mm of battery shift), **(c)** change in displacement of variable buoyancy drive (VBD) (red line; units of 10 cc), vertical velocity from pressure measurements (black line; cm s⁻¹), buoyancy (blue line; units of 10 g), **(d)** heading (desired heading: red line, measured heading: black line; degrees) **(e)** roll (battery roll position: red line, glider-measured roll: black line; degrees), and **(f)** vertical speed (calculated from buoyancy and pitch; black line; cm s⁻¹) and horizontal speed in centimeters per second (calculated from buoyancy and pitch; blue line; cm s⁻¹).

pare well with the discrete water samples (Fig. 8), overestimating the discrete water samples between 8.3 μatm (2.7 %) and 12.0 μatm (5.1 %) (Table 3). The mean difference between the rosette-mounted and Seaglider-integrated SG HydroC CO₂ sensors during the 7 May cast at the time of discrete samples was 8.5 $\mu\text{atm} \pm 8.9 \mu\text{atm}$ (3.7 %). The larger difference between SG HydroC CO₂ sensors compared to the difference during the tank experiment (see Sect. 3.2.1) is unsurprising, given the spatial and temporal distance between sensors (Table 3). Collecting more discrete samples throughout the water column and in closer proximity (within 100 m; Thompson et al., 2021) to the CO₂ Seaglider conducting dives would allow a more tightly constrained uncertainty

estimate for response-time-corrected $p\text{CO}_2$ data collected on a glider and should be a priority for future researchers.

3.3 CH₄ Seaglider data evaluation

Tank experiment

The SG HydroC CH₄ sensor was also evaluated during the tank experiment described in Sect. 2.4 (Fig. 6c). Percent differences (Eq. 1) between discrete $p\text{CH}_4$ (average of triplicate samples) and $p\text{CH}_4^{\text{RTC}}$ were 6.6 % to 15.8 % (Table 1). During the experiment, there was a decrease in salinity from 30.95 to 29.88, where $p\text{CO}_2$ decreased by 80 μatm . The corresponding $p\text{CH}_4^{\text{RTC}}$ signal decreased by 25.4 μatm from 32.3 to 6.9 μatm . Although the triplicate discrete $p\text{CH}_4$ water

samples were slightly lower than the sensor-measured $p\text{CH}_4$ values, they also reflected this step change.

3.4 Winter and springtime $p\text{CO}_2$ in Resurrection Bay, Alaska

The surface-to-subsurface $p\text{CO}_2$ gradient is much larger in spring than in winter (Fig. 11). During the 4–7 May mission, the average surface $p\text{CO}_2^{\text{RTC}}_{2,\text{Seaglider}}$ was $240.7 \pm 16.5 \mu\text{atm}$ (mean \pm standard deviation at 2 m) with an average temperature of $5.8 \pm 0.4^\circ\text{C}$ (Figs. 4 and 11). In February, surface $p\text{CO}_2^{\text{RTC}}_{2,\text{Seaglider}}$ was near atmospheric $p\text{CO}_2$ ($427.4 \pm 13.0 \mu\text{atm}$, temperature $4.1 \pm 0.3^\circ\text{C}$) and about $180 \mu\text{atm}$ higher than in May (Figs. 5 and 11). NOAA's moored sensor located in Sunny Cove (59.911°N , -149.35°W), near the CO_2 Seaglider trial site, measured an average sea surface $p\text{CO}_2$ of $240.7 \pm 10.4 \mu\text{atm}$ during the time of the May 2022 mission (Monacci et al., 2023), which compared remarkably well with the Seaglider-based measurements. A minimum of $140 \mu\text{atm}$ was measured in Sunny Cove in mid-April (3 d average) (Fig. 12; Monacci et al., 2023), suggesting that the peak of the spring bloom happened 3 weeks before the May 2022 glider mission. Since we do not have salinity data from the May CO_2 Seaglider mission (conductivity sensor failure), we cannot disentangle the contributions of freshwater or primary production to the low surface $p\text{CO}_2$ values observed (Fig. 4). The moored sensor in Sunny Cove measured an average sea surface $p\text{CO}_2$ of $416.4 \pm 4.2 \mu\text{atm}$ during the time of the February mission, straddling the atmospheric $p\text{CO}_2$ values (Monacci et al., 2023; Fig. 12). Subsurface $p\text{CO}_2^{\text{RTC}}_{2,\text{Seaglider}}$ at 180 m was on average $545.6 \pm 16.9 \mu\text{atm}$ during the February mission and $518.2 \pm 37.4 \mu\text{atm}$ during the May 2022 mission (Fig. 11a). $p\text{CO}_2$ was much lower in May than in February throughout the upper water column (< 120 m), whereas there was not much of a seasonal difference at deeper depth. Some of the fine-scale features apparent in the May $p\text{CO}_2$ and O_2 profiles are likely due to various levels of photosynthetic activity (Fig. 11). As the glider transitioned into the open Gulf of Alaska during the February mission, water with $\text{O}_2 < 150 \mu\text{M}$ shoaled into the upper 150 m of the water column (Fig. 5). Unfortunately, the HydroC CO_2 sensor was turned off at that stage of the mission to conserve battery.

4 Discussion

The newly developed CO_2 Seaglider is the first of its kind to autonomously collect high-quality $p\text{CO}_2$ data. The tank and rosette experiments and in situ data evaluation suggest that the post-processed data from the CO_2 Seaglider generally fall near the relative uncertainty of 2.5 %, which is a threshold defined as the “quality sufficient to identify relative spatial patterns and short-term variation” (“weather quality”; Newton et al., 2015). This is the highest quality of $p\text{CO}_2$ data

measured with a subsurface autonomous vehicle to date and therefore an important step towards filling the subsurface carbonate system data gap. -4H-JENA is reassessing their sensor calibration methodology and data post-processing algorithm to further improve the HydroC's data accuracy.

The newly developed CO_2 Seaglider is suitable for data collection in open-ocean or coastal environments with bottom depths deeper than 300 m. However, the coastal Gulf of Alaska is a highly dynamic environment, with strong freshwater and wind influence, as well as rugged shallow (often < 200 m) bottom topography. Strong currents (up to 0.50 m s^{-1}) made the piloting of the glider extremely difficult throughout the project and confirmed that the Seaglider cannot reliably reach desired waypoints in these conditions. The current version of the CO_2 Seaglider is also not suitable for operating in the coastal Gulf of Alaska in summer and early fall due to strong seasonal salinity gradients in this freshwater-influenced area. Another issue we faced was the fact that the forward-looking altimeter could not detect the seafloor as it should in its position behind the HydroC CO_2 sensor. In areas with detailed topography maps this would not be an issue, but in the coastal Gulf of Alaska reliable topography information is not readily available yet. An obvious next step is to integrate the SG HydroC CO_2 sensor into a newer glider platform, such as the Seaglider SGX or Teledyne Slocum G3 glider. The extended energy bay, larger buoyancy range, and thruster should make the operation of the coastal Slocum G3 with HydroC sensors relatively easy and would allow for autonomous high-resolution water column measurements of $p\text{CO}_2$ and $p\text{CH}_4$ in dynamic coastal environments. The integration of a HydroC on a Slocum glider will require a custom-made wet-payload bay due to the size of this sensor. For open-ocean or deeper coastal regions, the integration with the Seaglider SGX, with 60 % higher energy capacity, would be effective and nearly identical to the work already done here.

The SG HydroC CH_4 sensor was successfully integrated into the Seaglider as part of this project. While tank experiments showed promising results, short field tests of the CH_4 Seaglider in shallow water revealed low and patchy methane concentrations near the detection limit (not shown). The CH_4 Seaglider requires further testing in environments with strong $p\text{CH}_4$ gradients during longer and deeper dives (to allow for equilibration) to assess the accuracy of its response-time-corrected data in the field. The sensor's slow response time likely limits the glider to providing qualitative rather than quantitative results. However, due to the scarcity of oceanic CH_4 observations, deploying a CH_4 glider can help identify the location of methane sources and guide the placement of in situ observations to conduct a more quantitative assessment of CH_4 fluxes and dynamics.

Ocean gliders are part of the Intergovernmental Oceanographic Commission (IOC-UNESCO) Global Ocean Observing System (GOOS) through the OceanGliders program (<https://www.oceangliders.org/>, last access: 14 Oc-

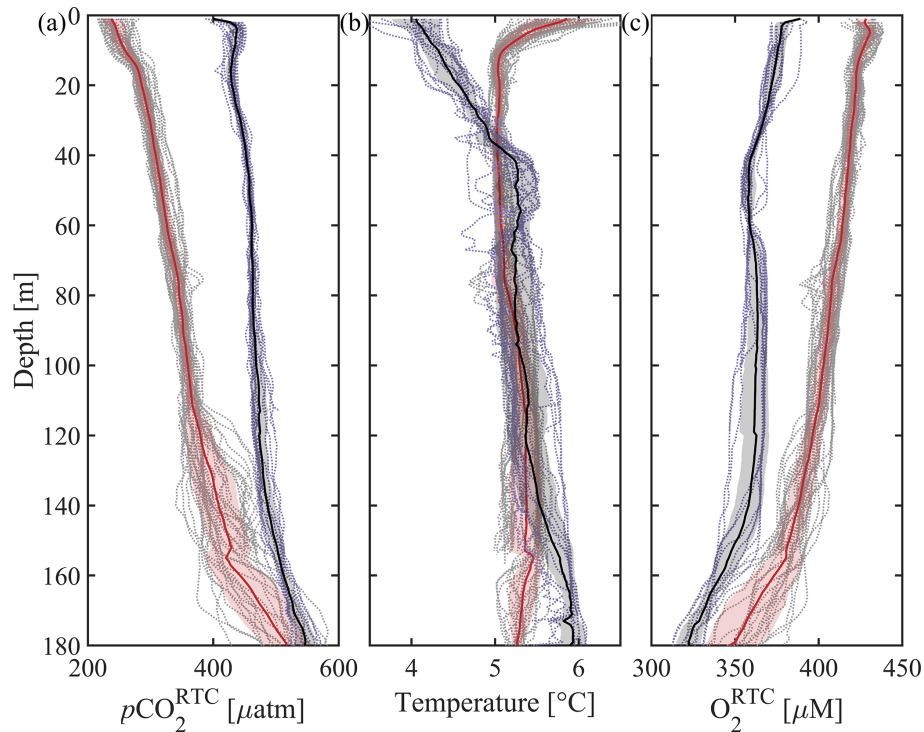


Figure 11. Averaged CO₂ Seaglider profiles from the 4–7 May 2022 and 8–21 February 2023 missions in Resurrection Bay, Seward, Alaska. Depth profiles of all 1 m binned dives (dotted gray) and the average 1 m binned dive from the May 2022 mission (thick red line; dives 1–51 at 00:01 UTC on 5 May to 16:37 UTC on 7 May 2022) and February 2023 mission (thick black line; dives 1–17 at 20:50 UTC on 8 February to 19:54 UTC on 9 February 2023), with shading showing the standard deviation of the values in each bin added and subtracted from the average. **(a)** Response-time-corrected $p\text{CO}_2$ ($p\text{CO}_2^{\text{RTC}}$, μatm), **(b)** temperature ($^{\circ}\text{C}$), and **(c)** response-time-corrected oxygen (O_2^{RTC} , μM).

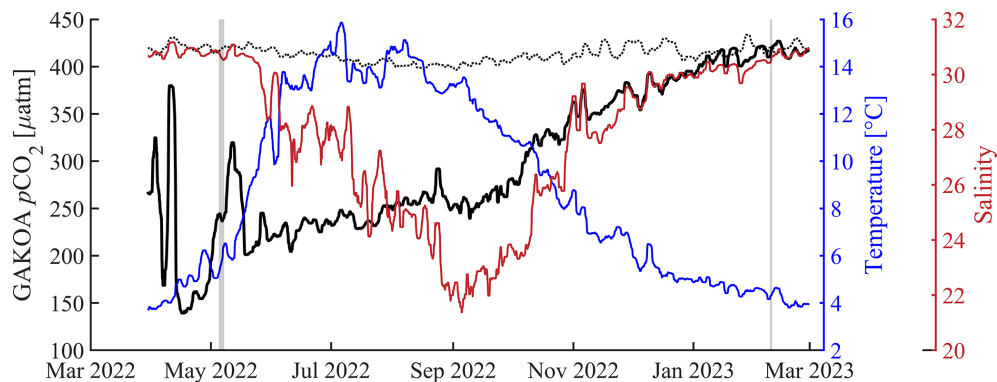


Figure 12. The National Oceanic Atmospheric Administration’s Gulf of Alaska ocean acidification surface time series from March 2022–2023. Left axis: sea surface (dotted black line) and air (black line, 4 m above sea level) $p\text{CO}_2$ (μatm); right axis: sea surface temperature (blue, $^{\circ}\text{C}$) and sea surface salinity (red). All data are shown as a 3 d running mean. Vertical shaded gray areas highlight the CO₂ Seaglider missions in May 2022 and February 2023. The mooring is located at 59.911° N, −149.35° W (Monacci et al., 2023).

tober 2024). Like other elements of the GOOS coordinated by OceanOPs of the Observation Coordination Group (floats, buoys, moorings, ships, and tide gauges), OceanGliders contributes to ocean observation for climate, ocean health, and real-time services. CO₂ gliders are perfectly suited to contribute data for understanding relevant inorganic carbon processes in coastal shelf and boundary re-

gions where mesoscale or sub-mesoscale variability dominates. The current work can also serve as a first step to bring together interested scientists and engineers to further develop and improve the capability of gliders to measure high-quality data. OceanGliders supports this effort by promoting the formation of volunteer international task teams, for which a task team could be requested for oceano-

graphic greenhouse gas research. By doing this, the visibility and availability of data will be improved as well, since GOOS provides an interactive data platform for all its programs (<https://www.ocean-ops.org/board>, last access: 14 October 2024). An OceanGliders task team could also be linked with the GOOS-sponsored Global Climate Observing System (GCOS: <https://gcos.wmo.int/en/home>, last access: 14 October 2024) program through their Ocean Observations Physics and Climate Panel (OOPC): a scientific expert advisory group charged with making recommendations for a sustained global ocean observing system for climate.

5 Concluding thoughts

Near-real-time and high-resolution water column data that can be retrieved from gliders outfitted with sensors measuring salinity, temperature, inorganic carbon system parameters, oxygen, and $p\text{CH}_4$ are key to tackling a variety of today's climate-change-related issues. These datasets will become instrumental in advancing biogeochemical model forecasting and early warning systems for extreme heat, acidity, and oxygen compound events that affect coastal subsistence communities, commercial fisheries, and mariculture. Furthermore, using biogeochemical gliders to monitor the environment of tagged organisms (e.g., crabs, fish) would provide insight into the organisms' position and behavior relative to important environmental drivers across susceptible ecosystems. Such biogeochemical glider data will help build a bridge between in situ chemical and biological measurements and environmental change with impacts on biology, thereby filling an important research gap (Widdicomb et al., 2023). Potentially large natural and anthropogenic sources of CH_4 may become contributors to climate change and, if oxidized, to ocean acidification (Garcia-Tigreros et al., 2021; Sparrow et al., 2018; Shakhova et al., 2010; Rees et al., 2022). These CH_4 sources need to be properly assessed and quantified, and if characterized as having anthropogenic origins, emitters must be held accountable (Goodman et al., 2022). Once the combined HydroC CH_4 – CO_2 sensor is available it will provide a new tool to co-measure $p\text{CH}_4$ and $p\text{CO}_2$ and give valuable insight into these processes and feedback mechanisms. Other advancing fields, such as marine carbon dioxide removal (mCDR) and the monitoring, verification, and reporting (MRV) thereof, will also need detailed knowledge of the distribution of CO_2 in the water column (National Academies of Sciences, Engineering, and Medicine, 2022).

The CO_2 Seaglider has been extensively tested and is ready to be used in open-ocean environments. An important next step will be to integrate the HydroC CO_2 and CH_4 sensors into a glider platform that reliably functions in shallow and freshwater-affected coastal areas, such as the Gulf of Alaska, to be able to fill the large spatial and temporal data gap in these highly dynamic areas.

Code and data availability. The CO_2 Seaglider data are publicly available at <https://doi.org/10.17882/100964> (Hauri et al., 2022) and <https://doi.org/10.17882/100965> (Hauri et al., 2023). The HydroC-specific SIRMA code and CNF file are available on GitHub at <https://github.com/Cyprus-Subsea/Smart-Cable-HydroC> (Cyprus Subsea, 2024a) and <https://github.com/Cyprus-Subsea/Smart-Cable-HydroC/tree/main/docs> (Cyprus Subsea, 2024b). More detailed information on the HydroC–glider integration and operation can be found in the CO_2 Seaglider Standard Operating Procedures at https://britairving.github.io/Carbon_Dioxide_SOP/README.html (Irving et al., 2024).

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Competing interests. Authors Dan Hayes and Ehsan Abdi are employed by AOOI and CSCS (respectively) and their objective is to support the ocean research community by providing innovative, cutting-edge observing technological solutions. These include autonomous platforms and related services in unique configurations. Through the support of the National Science Foundation and the National Oceanographic Partnership Program, AOOI was able to jointly develop the CO_2 and CH_4 gliders and prove and improve the scientific utility of this approach. Authors Nadja Kinski and Jöran Kemme are employed by -4H-JENA engineering GmbH, the manufacturer of the HydroC CO_2 and CH_4 sensors. The objective of -4H-JENA engineering GmbH is to provide the best possible accuracy of dissolved gas measurements on any platform and under any environmental conditions. Intensive collaboration with scientists is essential for the development of these products.

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